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## Drug Binding to Human Alpha-1-acid Glycoprotein in Health and Disease

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C. Structural and physical-chemical properties of the polypeptide and carbohydrate moiety of alpha-1-acid 

## I. Introduction

THE PLASMA binding of drugs can have important pharmacokinetic implications, especially when the drugs are highly bound having a binding constant larger than approximately  $10^5$  M<sup>-1</sup> and when their apparent volume of distribution is small (61, 136, 186, 266, 270, 307, 318, 347, 383, 391, 406, 446, 493, 497-499, 527, 537, 538, 573, 587).

Human serum albumin  $(HSA)^{\dagger}$  and alpha-1-acid glycoprotein (AGP) are the important drug binding proteins in plasma. HSA is the most abundant protein (4 g/100 ml of plasma), whereas the normal AGP level varies between about 50 and 100 mg/100 ml of plasma. The AGP level can vary considerably as a result of certain diseases, the use of drugs, and pregnancy. Values of up to 300 mg/100 ml of plasma have been found (396, 406,

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<sup>\*</sup>Abbreviations used are: AGP, alpha-1-acid glycoprotein; *B*, bound fraction of a drug; *B/F*, binding ratio of a drug;  $c_{\text{bound}}$ , bound concentration of a drug;  $c_{\text{tree}}$ , free concentration of a drug; *F*, free fraction of a drug;  $F_{AGP}$ , free fraction of a drug in an AGP solution;  $F_{HBA}$ , free fraction of a drug in an HSA solution;  $F_{AGP + HBA}$ , free fraction of a drug in a solution of a mixture of AGP and HSA; HSA, human serum albumin;  $K_{AGP}$ , drug-AGP binding constant;  $K_{HBA}$ , drug-HSA binding constant; LIPO, lipoprotein(s);  $n_{AGP}$ , number of binding sites on AGP;  $n_{HBA}$ , number of binding sites on HSA; *P*, plasma protein concentration;  $P_{AGP}$ , plasma concentration of AGP;  $P_{HBA}$ , plasma concentration of HSA ;  $P_{LIPO}$ , plasma concentration of LIPO. 426, 513, 538; see also table 6). HSA is largely responsible for the plasma binding of acidic drugs, whereas AGP binds mainly basic and neutral drugs. Although HSA has a greater binding capacity than AGP, especially for basic and neutral drugs, AGP can be the most important determinant in plasma binding, due to its greater drug affinity (44, 55, 75, 82, 83, 136, 194, 261, 358, 376, 385, 396, 450, 540, 544; see also table 8).

Drug monitoring is of increasing importance in clinical practice, especially in the case of drugs with a small therapeutic index. If such drugs are highly bound in the plasma and have a small volume of distribution, the free concentration of the drug in plasma will be a more reliable parameter for representing the intensity of the pharmacological effect than the total plasma concentration (3, 49, 71, 72, 79, 88, 189, 192, 202, 270, 277, 278, 308, 310, 321, 347, 395, 406, 408, 445, 518, 528, 538, 546, 556). If variations in the plasma levels of AGP occur, then the free plasma level of the drugs in question can vary considerably, whereas the total drug concentration of the drug in plasma will be only slightly affected (49, 108, 192, 210, 277, 278, 347).

A thorough review of the studies on the binding of drugs with AGP, both in vitro and in vivo, has been made in order to obtain a better understanding of the different factors which can affect the free plasma level of such drugs. These factors include the effects of exog-



enous and endogenous substances on the binding profile of drugs to AGP, the in vitro to in vivo correlations of free level determination, and the reliability of the free level determination. We have also studied what precautions should be taken in order to upgrade the reliability of the determination of the free concentration, and we have examined the role the physical-chemical properties of isolated samples of AGP play in results of in vitro binding studies. The binding of several therapeutic classes of drugs to AGP is reviewed. The value of the use of AGP as a diagnostic and prognostic acid in disease states is reviewed as well.

## II. Isolation, Structure, and Physical-Chemical Properties of Alpha-1-acid Glycoprotein

AGP, also called orosomucoid, has been a subject of study for more than 90 yr (257, 258, 396, 469). In table 1, a survey of these studies on AGP is given.

The isolation, the structure, and the physical-chemical properties of AGP have been reviewed earlier (257, 258, 469, 578). From these review studies it has become clear that there are several forms of AGP which differ in their structure and physical chemical properties. These forms have been described in terms of their physical-chemical properties (table 2).

Native AGP, asialo or desialylated AGP, modified AGP, and abnormal AGP are heterogeneous forms of AGP which differ in their molecular weight and/or electrophoretic pattern (tables 3 and 4). The molecular weight of native AGP, modified AGP, and abnormal AGP is about the same, whereas that of desialylated or asialo-AGP is lower (table 3). The amount of polymer that forms in native AGP during isolation determines the molecular weight of the polymers of AGP. The several heterogeneous forms of AGP have electrophoretic patterns which differ in the number of bands, in the moving velocities of these bands, and in the intensities of these bands, because of small charge differences in the peptide chain and carbohydrate moiety of AGP (table 4). AGP samples with different electrophoretic patterns are generally reported as different microheterogeneous types of AGP or simply as the occurring polymorphism of AGP (469). In the literature the names of heterogeneous forms or variants of AGP are sometimes, incorrectly, used to denote microheterogeneous types of AGP (469).

In this section the physical-chemical properties of several variants of AGP, such as molecular weight, stability, and microheterogeneity, will be reviewed.

## A. Methods for Isolating Alpha-1-acid Glycoprotein

Many studies dealing with methods for isolating AGP have been reported (54, 94, 98, 119, 173, 213, 219, 257, 258, 297, 298, 306, 319, 320, 334, 388, 469, 484, 560, 570, 579, 593, 594). All the procedures described are time consuming due to the use of a series of sequential chromatographic and/or precipitation steps. Recently twoand three-step purification methods, starting from Cohn Fraction VI, have been reported (217, 303, 505). Succari et al. (515) reported recently on a two-step purification method starting from plasma itself. Because in this procedure exposure to strongly acidic conditions was prevented, the investigators could obtain an AGP sample which had not undergone desialylation. Hellerstein et al. (225) recently described a time-saving isolation method

No.	Subject	Period	Ref.
I	Isolation and characterization	From 1882 until about the 1960s;	54, 98, 173, 257, 258, 297, 298, 334, 466, 468, 469, 476, 484, 485, 512, 560, 570, 579, 593, 594
		again from the end of the 1970s, due to the observation that the physical-chemical properties of AGP are dependent on the isola- tion procedures used	12, 23, 97, 102, 123, 217, 225, 303, 515; section II
п	AGP as acute phase protein	The 1960s and the 1970s	129, 137, 152, 196, 199, 293, 446; section III
III	AGP as drug carrier for ste- roids	The 1960s	181, 182, 565
IV	AGP as acute phase protein and as diagnostic and prog- nostic aid during therapy of several disease states	Since the 1980s	164, 165, 178, 184, 221, 472, 553; section III A
v	AGP as drug carrier, espe- cially for basic drugs;	Since the 1980s	396, 412, 460
	as drug carrier for some acidic drugs		249, 544, 545; section IV

# TABLE 1 Survey of the history of the studies on AGP

PHARMACOLOGICAL REVIEWS

		Survey of the several names use	d for AGP	
No.	Name	Origin name	Period used	Ref.
I	Tierisches Gummi	Carbohydrate substance, iso- lated from blood, with proper- ties identical to those of Tier- gummi (297), a mucoid iso- lated from snails	Sporadically, 1892	173, 297, 431
п	Seromucoid	Mucoid isolated from serum with properties comparable to those of ovomucoid, a mucoid isolated from eggs [ovum (Latin) = egg]	Always until about 1960, later sporadic- ally	27, 173, 221, 257, 258, 273, 431, 548, 578, 593, 594
ш	Alpha-1-acid glyco- protein	Acid plasma protein classified as an alpha-1-globulin and with a low isoelectric point (3.4) and a molecular weight of about 40,000	Since 1942 the name most often used	334, 475, and most refs. of table 6
IV	Mucoprotein	Glycoprotein with 30% to 50% carbohydrates	Very sporadically, about 1950	500, 560, 578, 579
v	Orosomucoid	Mucoid isolated from serum, with a high solubility in boil- ing water (98); [oros (Greek) = aqueous part of blood]	Since 1950 often used	53, 56, 57, 70, 93, 98, 199, 208, 211–213, 274, 302, 488, 560, 570
VI	Alpha-1-glycoprotein of Schultze	Alpha-1-glycoprotein with a mo- lecular weight of about 54,000, first described by Schultze et al. (484)	Very sporadically, 1955, 1962	152, 484
No.	Species	Defined as		Ref.
I	Native AGP	Isolated from plasma or serum, and same properties as in vivo	probably with the	53, 102, 107, 179, 213, 267, 268, 311, 371, 372, 388, 499, 500, 582
п	Delipidated or defat- ted AGP	AGP with a lower content of fatty a AGP, resulting from ethanolic or		97, 107, 181, 211, 212, 213, 282, 285
ш	Asiolo- or desialy- lated AGP	Native AGP from which essentially of the carbohydrate groups are re- but sometimes called modified AG	moved enzymatically;	20, 21, 50, 53, 73, 74, 102, 124, 177, 211, 268, 356, 421, 494, 499, 582
īV	Modified AGP	Native AGP from which amino acid chains are modified; also sometim AGP	53, 181, 259, 294, 335, 419, 489, 490, 499, 555	
v	Abnormal AGP	Native AGP, isolated from serum or with physical-chemical properties AGP isolated from serum of healt	73, 74, 104, 140, 162, 215, 250, 323, 392, 448, 489, 490, 595–598	
VI	Polymers of AGP	Polymers of native AGP formed due or purification; degree of polymer the procedures used		35, 212, 508, 569, 571
VII	Microheterogeneous types of AGP = polymorphism of AGP	Variants or heterogeneous forms of I–VI) with differences in their ele (different number of bands, differ bands, and/or different intensity	ctrophoretic patterns ent velocity of these	14, 15, 489, 490, 494, 529. see also refs. to table 4

 TABLE 2

 Surveys of the several names and variants of AGP

## TABLE 3 Survey of reported molecular weights for AGP

No.	Method used	Remarks on the AGP preparation used	Values	Ref.
I	Diffusion-viscosity	Native AGP, isolated according to Weimer et al. (560)	44,100	500
п	Light scattering	Two different forms of AGP, referred to as "alpha <sub>1</sub> -niedermolekulares Säureprotein" and "alpha <sub>1</sub> -Glykoprotein (3.5)" or "alpha <sub>1</sub> -acid	40,000	152, <b>483, 484</b>
	• J	glycoprotein of Schultze," respectively, iso- lated by using precipitation	54,000	152, 483, 484
	Adsorption methods			
ш	Sedimentation-diffu- sion	Native AGP, isolated from Cohn fraction VI by chromatography on carboxymethyl cellulose	41,600	53
	Sedimentation-viscosity	Same method	43,000	53
	Sedimentation-diffu- sion	Desialylated AGP, isolated as described above	38,600	53
	Sedimentation-viscosity	Same method	41,600	53
IV	Sedimentation-diffu- sion	Native AGP, isolated according to Weimer et al. (560)	37,700	267
	Sedimentation-viscosity	Native AGP, isolated according to Weimer et al. (560)	36,700	267
	Light scattering	Native AGP, isolated according to Weimer et al. (560)	48,000	267
v	Sodium dodecyl sulfate- polyacrylamide gel electrophoresis	Native AGP, isolated by electrofocusing	40,000	311
VI	Osmotic pressure, sedi-	Native AGP, isolated according to Buergi and	41,100	268
	mentation equilib-	Schmid (94)	39,000	268
	Osmotic pressure, sedi-	Desialylated AGP, isolated as above	34,600	268
	mentation equilib- rium		34,100	268
VII	Sedimentation-diffu- sion	Native AGP, isolated from plasma using several ion exchange chromatography methods	44,680	388
/111	Polyacrylamide alab gel electrophoresis	Two forms of AGP, one isolated from urine, the other from lymphocytes, granulocytes, and	41,000	180
	-	monocytes membranes	52,000	180
IX	Exclusion chromatogra- phy on Sephadex G-200	Two forms of AGP with common immunologi- cal determinants and almost identical amino	45,000	104
		acid composition but different amounts of car- bohydrate, isolated from liver metastases of sev- eral tumors	37,000	10 <del>4</del>
x	Not mentioned	Isolated by Behringwerke	44,100	100
XI	Polyacrylamide gel elec- trophoresis	Native AGP, isolated according to Gangula et al. (182); total carbohydrate content about 47%	45,000	371, 372
XII	Polyacrylamide slab gel electrophoresis	Native AGP, isolated according to Laurell et al. (300) followed by a purification using hydroxy- apatite high-pressure liquid chromatography	48,000	179



which is suitable for large numbers of small-volume samples of plasma. They pointed out that it was important to check for the possible occurrence of desialylation during the acid precipitations. Halsall et al. (213) described an isolation method for native AGP from nephrotic urine under practically physiological experimental conditions. Arnaud et al. (23) described a preparative isoelectric focusing procedure starting from albumindepleted serum, which resulted in the separation of at least seven microheterogeneous types of AGP (section II D). Downloaded from pharmrev.aspetjournals.org at Thammasart University on December 8, 2012

On studying the literature dealing with AGP, one notices that the stability, denaturation, and polymerization of AGP are hardly discussed at all in clinical, phar-

No.	AGP used and method used	Results	Ref.
I	Neuraminidase-treated AGP; starch-gel electrophoresis, pH 4.8	Three types, each having two bands: Type I Slower moving main band and faster moving minor band Type II Faster moving main band and slower moving minor band Type III Two main bands at same position as main bands of types I and II	535
	Native AGP; starch-gel elec- trophoresis, pH 2.9	Three types with 7, 6, and 7 bands, respectively, and corresponding to the types I, II, and III, respectively, but after neuraminidase treat- ment of these three native types	535
П	Native AGP; starch-gel elec- trophoresis, pH 2.9	Four types with 5, 6, 7, and 8 bands, respec- tively, occurring at a relative incidence of 4, 36, 49, and 11%, respectively, and probably due to genetically determined types	470
ш	Neuraminidase-treated AGP; starch-gel electrophoresis, pH 5.1	Three types, each with two bands; relative per- centages of these three types differ between a white and a Japanese population; types ge- netically determined; different types due to differences in polypeptide moiety	478
IV	Neuraminidase-treated AGP; starch-gel electrophoresis, pH 5	Three types, each with two bands; types inde- pendent of stress (after surgery, during preg- nancy, and after delivery), and genetically de- termined	534
v	Neuraminidase-treated AGP from plasma of patients with uterectomy and irra- distion; starch-gel electro- phoresis, pH 5	Three types, each with two bands; types inde- pendent of disease and AGP plasma level; types genetically determined	591
VI	Neuraminidase-treated whole serum; agarose-gel electrophoresis, pH 5, and immunofixation	Three types, each with two bands, called SS, FE, and FS, corresponding with types I, II, and III, respectively; types genetically deter- mined; types due to differences of amino acid composition of the peptide chain resulting in F and S bands with different electrophoretic mobilities	263
VII	Native AGP; isoelectric fo- cusing	Two types with a relatively anodic and cathodic distribution of 6 to 8 bands, respectively; isoelectric points range from 2.90 to 3.30	52
	Neuraminidase-treated AGP; isoelectric focusing	Two types with one or two main bands, both exhibiting several minor components; isoelec- tric points, 4.55 and 4.70, respectively; pat- tern not correlated with those of native AGP; microheterogeneity due to amino acid re- placements of polypeptide chain in combina- tion with different linkages of sialic acid to carbohydrate residues in native AGP	52
VIII	Native AGP; isoelectric fo- cusing and titration curves	At least seven bands with isoelectric points be- tween 3.4 and 3.8; microheterogeneity very slight, between pH 6 and 8	23
	Neuraminidase-treated AGP; isoelectric focusing and ti- tration curves	The same pattern as for native AGP, but with isoelectric points between 4.3 and 4.7; very alight microheterogeneity, between pH 6 and 8; microheterogeneity not due to differences in sialylation, but to other mechanisms	23

## TABLE 4 esults of reported microheterogeneous studies of AGP

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No.	AGP used and method used	Results	Ref.
IX	Whole plasma of depressive patients; isoelectric focus- ing	Three types with 6, 7, and 8 bands, respectively, independent of total AGP level; isoelectric points range from 3.2 to 3.9; types due to genetically determined variants	530
x	Sera of cancer patients; crossed-immunoaffinity electrophoresis; binding to wheat germ agglutinin	Distribution of three bands changed in cancer disease; lower binding in cancer disease to wheat germ agglutinin possibly due to dimin- ished content of sialic acids in outer part of carbohydrate moiety	65
XI	Crossed-immunoaffinity electrophoresis; influence of estrogen level	Pattern with three bands, changing to a pattern with two bands with more of the concana- valin A nonreactive bands after increase of sex hormone levels (during pregnancy and after estrogen therapy of prostatic cancer)	563
XII	Sera of healthy people, of cancer patients, and of women during pregnancy; crossed-immunoaffinity electrophoresis	Three bands in serum of normal subjects, but only two bands in serum of women during pregnancy and in serum of prostatic cancer patients treated with estrogen; increase of faster moving bands and disappearance of concanavalin A reactive band	427
XIII	Normal and inflammatory sera; crossed-immunoaf- finity electrophoresis	Three bands with differences in pattern of dis- tribution between normal and inflammatory sera; increase of concavalin A reactive and concanavalin A weakly reactive bands during inflammation	373
	Concanavalin A affinity chromatography followed by isoelectric focusing	Only two bands, when separated by chromatog- raphy, namely a concanavalin A nonreactive band, which after isoelectrofocusing had 6 bands between pH 2.9 and 3.1 and 3 bands between pH 3.1 and 3.4, and a concanavalin A reactive band with 6 bands between pH 3.1 and 3.4 when followed by isoelectric focusing	374
	Concanavalin A affinity chromatography followed by crossed-immunoaffinity electrophoresis	Only two bands, when separated by chromatog- raphy; the concanavalin A nonreactive bands separated chromatographically contain, when followed by crossed-immunoaffinity electro- phoresis, the nonreactive and the weakly reactive bands, whereas the A reactive com- ponent separated chromatographically con- tained, after crossed-immunoaffinity electro- phoresis, a little weakly reactive band too; this band is also present in sera after crossed- immunoaffinity electrophoresis alone	374
XIV	AGP from sera of normals and patients with neoplas- tic disease; crossed-immu- noaffinity electrophoresis followed by isoelectric fo- cusing	In neoplastic disease additional bands between pH 3.7 and 4.4, compared with normals hav- ing bands between pH 3.2 and 3.8; due to differences in amino acid substitution and the presence of a non-covalently bound chromophoric group	598
xv	Crossed-immunoaffinity electrophoresis	Patterns with three bands, but with variable distribution of these bands; in severe disease and pregnancy these patterns change towards the less concanavalin A binding bands due to changes in the glycosylation of carbohydrate moiety of AGP, depending on severity of dis- ease state	68, 69
XVI	Crossed-immunoaffinity electrophoresis	Three bands with a relative distribution of 44.5%, 40.4%, and 16.1%, respectively, being constant under nonpathological conditions	237

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## DRUG BINDING TO HUMAN AGP IN HEALTH AND DISEASE

No.	AGP used and method used	Results	Ref.
хчи	Crossed-immunoaffinity electrophoresis	Four bands being differently distributed in nor- mal health, inflammatory lung disease, and cancer of the lung; benign sera contain more of the concanavalin A nonreactive bands whereas cancer sera contain more of the con- canavalin A reactive band; aid in diagnosis of cancer	214, 216

macological, or physical-chemical studies, although they are mentioned briefly in a few more fundamental physical studies (35, 97, 211–213, 274, 388, 507, 508). Presumably, most researchers assume that AGP is a very stable plasma protein. However, it should be noted that a temperature-dependent polymerization of AGP has been described, yielding two kinds of polymers differing in their biological activity (35) and in their drug-binding behavior (508). Halsall and Kirley (211) observed a temperature-dependent denaturation of AGP which is influenced by the degree of defatting and desialylation. From these observations, it can be concluded that sterilization of AGP by heating can induce denaturation of AGP.

Halsall et al. (212) reported (a) that aggregates of AGP are formed as a result of lyophilization or ultrafiltration; (b) that the acid-charcoal defatting procedure of Chen (109) induced polymer formation of AGP; (c) that lyophilization, especially of defatted AGP, induced polymerization, whereas the extent of polymerization proved to be dependent on the medium, the number of lyophylization cycles, and the protein concentration (increasing polymerization with decreasing protein concentration). They suggested therefore that care must be taken with defatted AGP and that lyophilization must be performed after extensive dialysis of an aqueous solution. Halsall et al. (212) observed further that AGP could be stored for 1 wk at 4°C in phosphate-buffered saline solution without the occurrence of polymerization, and that repeated freezing of an AGP solution and subsequent thawing did not result in polymerization. Recently, Busby and Ingham (97) reported that the thermal stability of AGP (as determined with fluorescent probes) is enhanced by lipids, propranolol, ethanol, and probably other organic solvents.

These studies demonstrate that different isolation and purification procedures result in AGP preparations with different physical-chemical properties.

### B. Molecular Weight of Alpha-1-acid Glycoprotein

The molecular weights reported for AGP (see table 3) range from 37,000 to 54,000. These values depend on the methods of determination (53, 268), on the isolation procedure (97, 104, 152, 180, 205, 484), on whether the AGP is native or desialylated (e.g., as a result of neuraminidase treatment; 53, 268), and on the origin of the AGP samples (from plasma, urine, or membranes of normals or patients; 104, 180, 213, 215, 323).

The molecular weight generally assumed for AGP is 40,000, which is about the mean value of the molecular weights reported for native AGP isolated from plasma (reported values, 38,800 to 48,000; table 3). Higher molecular weights for AGP have been reported by Schultze et al. (484; table 3, no. II), Easton et al. (152), Hardwick and de Vaux St. Cyr (219), and Gahmberg and Andersson (180; table 3, no. VIII). Gahmberg and Andersson (180; table 3, no. VIII) reported, however, that the AGP with a molecular weight of 52,000 was a membrane-bound form of AGP, synthesized by the lymphocytes, but subsequently cleaved and released as the soluble serum form of AGP with a molecular weight of 41,000. Hardwick and de Vaux St. Cyr (219) and Easton et al. (152) reported the isolation from urine and serum of two AGP variants. with a molecular weight of 40,000 and 54,000, respectively. From table 3 it follows that desialylated AGP has a mean molecular weight of about 38,000 (reported values between 34,100 and 41,600; table 3).

Recently the amino acid sequence of human AGP has been inferred from the cDNA sequence using the molecular cloning technique (64). The molecular weight of the polypeptide moiety studied can be easily calculated from this sequence. Board et al. (64) remarked, however, that clones with different sequences can not yet be excluded. In order to find the total molecular weight of human AGP, the molecular weight of the five glycan chains should be added to that of the polypeptide chain.

## C. Structural and Physical-Chemical Properties of the Polypeptide and Carbohydrate Moiety of Alpha-1-acid Glycoprotein

The chemical properties of AGP have been reviewed by Jeanloz (256, 257) and Schmid (469). AGP contains carbohydrate residues chemically bound to the protein. Therefore it can be catalogued among the groups of the glycoproteins (279, 487, 512), the mucoproteins (512, 560), the seroglobulins (298), and the alpha-1-globulins (265, 484).

AGP is composed of a single polypeptide chain and five carbohydrate moieties. Recently it has been shown that the polypeptide chain consists of 183 amino acids (64) [in contrast to the number of 181 reported earlier (469)] and contains two disulfide bonds (469, 471). The complete amino acid sequence, the multiple amino acid substitutions (21 of the 181 residues), and the homology with the immunoglobulins (about 80%) have been re-

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ported by Schmid et al. (474, 475). Recently the amino acid sequence of the polypeptide chain of AGP was deduced from the cDNA nucleotide sequence (64). There was an excellent agreement with the amino acid sequence reported earlier by Schmid et al. (475). There was only a difference at four places on the polypeptide chain (64, 469, 475).

The five carbohydrate moieties of AGP are located in the first half of the peptide chain and are linked to asparagine residues. The carbohydrate moieties consist of about 11% sialic acid, 14% neutral hexoses, 14% hexosamine, and 1% fructose (512). It should be pointed out here that human plasma proteins contain only Nacetylneuraminic acid, whereas proteins of other species may have variable proportions of the N-acetyl and Nglycosyl derivatives (382, 512, 577). The sialic acid residues, being easily removable (224), may be linked to C-2, C-3, C-4, or C-6 of the galactose residues (258, 469). The unusually low isoionic point of 3.4 is caused by the high sialic acid content (257, 469, 557). The literature up to 1972 dealing with the chemical identification of each of the five different carbohydrate moieties of AGP is reviewed by Jeanloz (257). The structure of the carbohydrate moiety of AGP has been studied extensively in recent years (25, 167, 224, 296, 477, 486, 487, 590). Fournet et al. (167) determined the primary structures of 16 asialo carbohydrate units derived from AGP, using 360 MHz proton nuclear magnetic resonance (NMR) spectroscopy. The asialocarbohydrate units can be grouped in five classes with bi-, tri-, and tetraantennary structures, respectively, for the first three classes. The fourth and fifth classes have also a tri- or a tetraantennary structure, but with an additional fucose residue. In addition to the five chains reported by Fournet et al. (167), Yoshima et al. (590) elucidated the structure of three new sugar chains. Hansen et al. (215) reported recently significant differences in antennary structure of the glycan part of AGP from different patient groups.

Recently, Cardon et al. (102) succeeded in analyzing the sialyloligosaccharides of AGP by high-performance liquid chromatography. They found that native AGP contains no traces of neutral oligosaccharides, but only monosialylated (5.8%), disialylated (34.6%), trisialylated (43.3%), and tetrasialylated (16.2%) glycans.

Crystals of AGP have been described (344, 345, 467). However, a detailed three-dimensional structure as determined from X-ray crystallography has not yet been reported. Schmid et al. (473) studied the tertiary structure of AGP in solution using circular dichroism (259) and chemical modification methods.

Aubert and Loucheux-Lefebvre (25) reported that the protein moiety of AGP contains 21% alpha-helix, 21% beta-sheet, 8 reverse beta-turns, and 40% unordered structure. They observed that, of the five carbohydrate moieties, four are linked to asparagine residues which are situated either in a reverse beta-turn or in regions where charged and polar residues are numerous, that is, on the outside of the protein. They also reported that the carbohydrate moieties do not produce any perturbation of the protein conformation. Schmid et al. (472) reported that, even after removal of 85% of the carbohydrate content, the secondary structure of the AGP was not affected.

Several studies show that the physical-chemical properties of AGP can change during disease states. Recently, Chandrasekaran et al. (104) isolated from liver metastases of lung, colon, and breast tumors two variants of AGP with common immunological determinants and almost identical amino acid compositions but different amounts of carbohydrate. Rudman et al. (448) found an abnormal AGP in the plasma of patients with neoplastic disease. This abnormal AGP had a molecular weight of between 40,000 and 50,000, a normal protein moiety, but multiple abnormalities in the carbohydrate moiety. An AGP variant in the plasma of cancer patients has been reported (162, 595-598). This abnormal AGP contains a chromophoric group which has the characteristics of a pteridine. It has a less negative charge, although its sialic acid content is not reduced. It exists partially in a polymeric form, possibly due to the cross-linking effect of the chromophoric group. Its extinction values and optical rotary dispersion (ORD) data indicate differences in the secondary structure. Ziegler et al. (596, 597) reported recently that the decreased negative charge of this AGP (595) is due to the binding of the pteridine chromophore to the sialic acid antennae. Serbource-Goguel et al. (489, 490) reported the presence of partially desialylated AGP in plasma from patients with liver disease. The degree of desialylation of AGP was dependent on the severity of the liver disease.

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### D. Microheterogeneity of Alpha-1-acid Glycoprotein

The several heterogeneous forms or variants of AGP (table 2) can have different electrophoretic patterns. This phenomenon is called the microheterogeneity of AGP.

The results of the studies on the microheterogeneity of the variants of AGP are summarized in table 4. From this table it follows that the microheterogeneity is dependent on the state of AGP (native or asialo), the characterization technique used, and the origin of the AGP preparation (from normal volunteers or patients; 14, 62, 73, 74). Umetsu et al. (543) introduced recently a new technique for isoelectric focusing, which is not included in table 4. Hanssen et al. (215) reported recently that the electrophoretic microheterogeneity of AGP can be evaluated in terms of the antennary structure of the glycan part of AGP and that significant differences in glycan structure were found in different patient groups. Serbource-Goguel et al. (489, 490) reported recently about the alterations in relative proportions of microheterogeneous forms of AGP in liver disease. Mackiewicz et al. (323) observed that the microheterogeneous forms can be used as indicators of rheumatoid arthritis activity.

Charge differences in the polypeptide and carbohy-

drate chains and structural differences in the carbohydrate moiety of AGP also play a role in the observed microheterogeneity: differences in the polypeptide chain are determined genetically, whereas differences in the carbohydrate moiety are dependent on the severity of the disease (table 4).

Tinguely et al. (529) reported that the S-variant (see table 4, no. VI) of AGP has a somewhat stronger affinity for amitriptyline and nortriptyline (table 8, nos. XXXII and III). Up till now, no other studies have been reported on the effect of the microheterogeneity of AGP on the pharmacokinetics of the drug binding (15).

## III. Biological Functions of Alpha-1-acid Glycoprotein

### A. Alpha-1-acid Glycoprotein as Acute Phase Protein

Since the sixties it has become clear that AGP is a plasma protein, the level of which can vary considerably during several physiological and pathological conditions. Tables 5 and 6 give a survey of these conditions. The variations in the AGP level proved to be dependent on the severity of the disease states. Whereas for healthy people plasma levels of AGP are reported to range between about 40 and 110 mg/100 ml, AGP values of up to about 300 mg/100 ml have been found during diseases (396, 406, 426, 513, 538). In order to substantiate these conclusions, data were collected from the literature. Table 6 gives a survey of quantitative data on AGP and HSA levels in the plasma of healthy people and of patients with various diseases.

From the data in table 6, it follows that the normal value of the average HSA concentration in plasma of about 4 g/100 ml can decrease until about 2 g/100 ml during disease. It further follows that the normal average plasma levels of AGP are between 50 and 100 mg/100 ml; 65% of the normal cases have a level between 60 and 80 mg/100 ml; the average value is 73 mg/100 ml. The table also gives data concerning the increase in AGP concentration in acute phase situations. About 50% of the data represent a situation in which the average value in the acute phase is twice as high as the average value in the normal situation. About 35% of the data give values that are 3 times as high. It can therefore be concluded that both AGP and HSA can be classified among the acute phase proteins (12, 293) and that especially the level of AGP in plasma can be used as diagnostic and prognostic aid during the treatment of several diseases (12, 164, 165, 178, 184, 216, 218, 221, 333, 351, 392, 423, 434, 496, 533, 541, 553, 554, 562, 595). In this study, only the acute phase behavior of AGP will be discussed extensively. The observed increased AGP level and decreased HSA level are also important in relation to drug binding; this will be discussed in section IV.

Other reports are known in which the AGP concentration was measured as a function of time (10, 80, 105, 106, 114, 122, 129, 133, 137, 152, 188, 196, 198, 199, 216, 218, 240, 260, 288, 323, 446, 454, 525, 562, 586). Monitoring

 TABLE 5

 Survey of several disease states and physiological conditions with varying AGP levels in human plasma

	varying AGP levels in human	n plasma
No.	Pathological/physiological condition	Ref.
I	Acidosis	140, 343
п	Age	5, 6, 16, 59, 60, 70, 77,
		134, 139, 201, 242,
		287, 290, 391, 397,
		398; table 6, nos.
TTT	Alashalasa	III-VII, L, LI
III IV		16, 26, 190, 452
V	Allergy (Ventricular) arrhythmia	103, 295, 393 10a, 161
vi	Arthritis	1, <b>42–44</b> , 139, 288,
*1	AIWIIIB	323, 409, 479–481,
		524
VII	Bacterial infection in	13, 56, 57, 70, 183,
	neonatal period	454, 455; table 6,
	-	nos. VIII–XII
VIII	Burn	63, 332; table 6, nos.
		XIV, XV
IX	Cancer (breast, colorec-	7-9, 76, 110-112, 114,
	tal, lung, ovaries)	116, 162, 164, 184,
		186, 215, 221, 241,
		250, 273, 392, 409,
		434, 502, 523, 526,
		541, 554, 559, 562;
		table 6, nos. XVI– XXXV
х	Chest pain	80, 169, 269, 400; ta-
л	Cheet pain	ble 6, no. LXXII
XI	Chronic inactive pyelo-	426
	nephritis	
ХП	Chronic hemodialysis	147, 226, 290, 391,
	patients	435, 589
XIII	Chronic pain	178; table 6, no.
	-	LXXXII
XIV	Chronic renal failure	146, 147, 435
XV	Chronic ulcerative colitis	137
XVI	Crohn's disease	152, 409, 479, 481
XVII	Depression	85, 86, 192, 193, 380,
		552; table 6, nos.
vvm	Failman	XLV, XLVI
XVIII	Epilepsy	304, 339, 432, 441, 531; table 6, nos.
		XLVII-IL
XIX	Genetic factor	16, 59, 60; table 6, no.
		LII
XX	Gliomas	562
XXI	Hepatitis	391, 489, 490; table 6,
	-	nos. LV, LVI
XXII	Hormonal contraceptives	59, 81, 301, 302, 408,
	use	585; table 6, nos.
		LXXX, LXXXI
XXIII	Hyperlipoproteinemia	145, 147, 226
XXIV	Hyperlipidemia Hyperlipidemia	<b>99</b> , 153, 154
	Hypertension Inflammation	200 62 137 253 271 409 -
XXVI	mianination	62, 137, 253, 271, 409, <sup></sup> 423, 430, 442, 479,
		480-482, 488, 561;
		table 6, nos. LVIII,
		LXIII
XXVII	Liver cirrhosis	26, 36, 187, 190, 391,
		409, 447, 489, 490,
		493, 524; table 6,
		nos. XXXVIII-
		XLI

9

**a**spet

TABLE 5—Continued

No.	Pathological/physiological condition	Ref.
XXVIII	Liver carcinoma	114
XXIX	Multiple sclerosis	426
XXX	Myocardial infarction	10, 34, 80, 105, 106,
		133, 135, 156, 161,
		169, 260, 269, 400,
		439, 440, 443, 447,
		449, 493, 504, 553,
		583; table 6, nos.
		LXVIII-LXXVII
XXXI	Nephrotic disease	145, 147, 391, 496,
		509, 589
XXXII	Obesity	16, 47, 48; table 6,
		nos. LXXVIII-
		LXXIX
XXXIII	Pregnancy	70, 117, 128, 135, 183,
		204, 218, 235, 236,
		265, 357, 367, 403,
		404, 427, 454, 455,
		521, 585; table 6,
		nos. LXXXIV-
		XCIII
XXXIV	Renal disease	120, 123, 145, 146,
		206, 270, 391, 409,
		435, 447, 493, 524;
		table 6, nos. XCIV,
		XCV
XXXV	Sex	5, 47, 59, 70, 139, 152,
		163, 178, 193, 287,
		290, 380, 442; table
		6, nos. LXXI,
		LXXXIX
XXXVI	Smoking	46, 59, 134, 241, 287;
		table 6, nos.
		XCVII, XCVIII
XXXVII	Stress	156, 178, 534
XXXVIII	Surgery	129, 135, 152, 163,
		171, 215, 222, 240,
	_	351, 409, 553
XXXIX	Trauma	129, 153–155; table 6,
274	<b>TT</b>	nos. IC-CVI
XL	Uremic disease	5, 145, 147, 206, 226;
		table 6, nos. CVII-
VII	Wound book	CXI
XLI	Wound healing	198, 351

AGP levels in this way is a useful aid in clinical therapy (table 6). Most of these data are collected from cancer patients. The elevated level of AGP for these patients (but not as a function of time) has been described extensively (76, 116, 221, 241, 273, 333, 434, 502, 523, 526, 541, 554, 559, 562; table 6, nos. XVI-XXXV). Changes in the AGP level have been correlated with the response of cancer patients to chemotherapy treatment (164, 184, 221, 591).

After myocardial infarction, higher levels of AGP (Table 6, nos. LXVII-LXXVII) with peak values on days 4 to 5 have been reported (10, 260), although Voulgari et al. (553) could not detect an appreciable change in the AGP level during the first 10 days after myocardial infarction. Other reports substantiate the use of AGP levels for diagnostic and prognostic purposes in this field (80, 105, 106, 504). Rises in the level of AGP (for survey, see table 6) have also been observed after surgery (129, 152, 163, 351, 553), in inflammation (10, 137, 351), and during infections (56, 57, 117, 199, 279, 405, 453–455, 553). Other applications have been described in patients with chronic pain (178), rheumatoid arthritis (323), hepatic diseases (114), multiple sclerosis (426), renal dysfunction (146), and during wound healing (198, 351).

AGP levels were also monitored during pregnancy (table 6, nos. LXXXIV-XCIII). A decrease depending on the stage of gestation was observed (57, 70, 196, 236, 506, 521). Levels of AGP have been reported to be higher in the first and third trimester with a decline around 24 wk gestation (218). The use of contraceptive steroids also decreases the AGP level (59, 81, 301, 302, 408, 585; table 6, nos. LXXX-LXXXI).

Lower AGP levels were observed in the serum of patients with liver cirrhosis (26, 36, 187, 190, 391, 409, 447, 489, 490, 493, 524; table 6, nos. XXXIX-XLI) and in the serum of newborns (13, 57, 70, 183, 199, 284, 404, 423, 554, 555, 585; table 6, nos. IX-XII, LXXXVII-XCIII). The occurrence of lower AGP levels in serum of newborns (more than 2 times less than the value in healthy adults) can explain the complications that occur directly after delivery in a mother using a drug therapy.

It would be interesting to discover the reasons for this changing level of AGP. Winzler and Bocci (578) reviewed the turnover of the major plasma glycoproteins. They reported that most of the circulating plasma glycoproteins, including AGP, are synthesized in the liver (51, 127, 262, 352, 355, 456, 580), probably in the form of an intrahepatic precursor (366). Weisman et al. (561) studied the turnover of AGP in man and observed an increase in synthesis in patients with inflammatory disease. Several reports in which the mechanism of the synthesis of AGP was studied using a perfused rat liver or cultures of rat hepatocytes (38, 144) point to the role of increased mRNA in this context. Diarra-Mehrpour et al. (144) recently reported that, after a high dose of 17-alphaethynylestradiol and after acute inflammation, rats showed an increase in the plasma concentration of AGP, due to hepatic accumulation of the AGP-mRNA. They further concluded that different mechanisms and/or pathways are probably involved in regulating the synthesis of AGP under various stimuli such as glucocorticoids. Similar conclusions were reached by other investigators (141, 291, 312, 428, 430, 547).

The homology with immunoglobulins is also stressed. Ikenaka et al. (248) reported that AGP was the first single-chain protein that was found to show sequence similarity with haptoglobin and particularly with the immunoglobulins (36% and 75%, respectively). Schmid et al. also observed the homology of AGP with the immunoglobulins (474, 475).

Toh et al. (532) reported recently that membrane AGP has some structural homologies with the  $\beta$ -chain of HLA-DC, with immunoglobulin, and with the epidermal

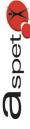
## TABLE 6

Survey of AGP and HSA levels in plasma of healthy people and patients with various diseases as reported in studies dealing with acute phase

		Disease level		Normal lev	vel	
No.	Disease	AGP [mg/100 ml]	HSA [g/100 ml]	AGP [mg/100 ml]	HSA [g/100 ml]	Ref.
I	Healthy people			70-110	3.9-5.5	513
			Two populations	$4.7 \pm 0.5$		287
			$105 \pm 31$ 64 ± 13	$4.7 \pm 0.3$		
II	Atherosclerosis	$115 \pm 16.9$ but decrease		$60 \pm 20$		99
		to 90 ± 24.8 after clofibrate treatment				
ш	Arthritis	$149 \pm 60$	$2.9 \pm 0.8$	66 ± 30	$4.1 \pm 0.8$	409
IV	Arthritis	$131 \pm 42$		<b>63 ± 17</b>		436
				$57 \pm 13$ (women) $69 \pm 19$ (men)		
v	Arthritis	$180 \pm 80$	$4.01 \pm 0.8$	$104 \pm 33$ (age dependent)	4.3 ± 0.7	139
VI	Arthritis	$213 \pm 42$		$64 \pm 16$		427
VII	Arthritis	$40 \pm 20$		$25 \pm 5$		185
VIII	Bacterial infection in neonates	Increase from 32–67 on day 3		32		13
IX	<b>Bacterial</b> infection	Increase to about 150		Increase from 17 $\pm$		57
	in neonates	$\pm$ 50, whereas during		3 at birth to 50 $\pm$		
		viral or parasite in- fection, increase to		10 at day 6		
v	D	about $100 \pm 50$ only		T		100
х	Bacterial infec- tions in neo- nates	Increase to 150		Increase from $8 \pm 4$ at birth to $27 \pm 4$		199
XI	Bacterial infec-	$133 \pm 75$ for neonates		after 7 days Increase from $18 \pm$		454
Л	tions in neo-	with favorable out-		8 at birth to 52 $\pm$		454
	nates	come, but $167 \pm 67$ for neonates with poor clinical outcome		8 after 2 days		
XII	Bacterial infec- tions in neo-	Increase to about 140		$30 \pm 25$ at birth and on day 4		70
	nates			about 50 ± 25		
XIII	Bronchitis	$30 \pm 10$		$25 \pm 5$		185
XIV	Burn	Increase to 222	Decrease to 2.2	83	4.4	63
XV	Burn injury	Increase to about 268 and 221 between days 5 and 25, re- spectively	Decrease to 1.9– 2.7 between days 8 and 25	66 ± 30	3 ± 1	332
XVI	Cancer of lungs	$217 \pm 29$		<b>76 ± 5</b>		9
XVII	Cancer of lungs, stomach, pan- creas, uterus, breast	270 ± 60		99 ± 8.3		27
XVIII	Cancer of breast, colon	118 ± 72	$3.47 \pm 0.61$	75 ± 18	3.95 ± 0.5	76
XIX	Cancer of liver	$148 \pm 53$		70.3 ± 2.54		114
XX	Cancer of colon	Increase to $128 \pm 62$ , depending on stage		$65 \pm 40$		116
XXI	Ovarian cancer	Increase to 240 de- pending on stage and effect of therapy		76 ± 14		164
XXII	Cancer of breast	Increase to 233		$76 \pm 14$		165
XXIII	Cancer	Increase to $54 \pm 22$ depending on severity		32 ± 5		185
XXIV	Active lung cancer	$120 \pm 36$ , whereas in inactive lung cancer, only $60 \pm 24$		60 ± 24		184
XXV	Cancer of lung	$216 \pm 20$ , but in lung inflammation, only $176 \pm 24$		76 ± 4		216

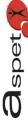
		TABLE 6—Continued Disease level		Normal level		
No.	Disease	AGP [mg/100 ml]	HSA [g/100 ml]	AGP [mg/100 ml]	HSA [g/100 ml]	Ref.
XXVI	Cancer	Increase to $137 \pm 20$ depending on stage and effect of therapy		73 ± 13		221
XXVII	Cancer	Increase to $352 \pm 163$ depending on sever- ity		92 ± 25		241
XXVIII	Cancer	$134 \pm 49$		$53 \pm 11$		321 <b>a</b>
XXIX	Prostatic cancer + stilbestrol	61 ± 7		64 ± 16		427
XXX	Advanced cancer	$238 \pm 71$		$64 \pm 16$		427
XXXI	Cancer	$255 \pm 41$		69 ± 8		502
XXXII	Prostatic cancer	Increase to about 126 $\pm$ 50, but 54 $\pm$ 29 during stilbestrol therapy		Decrease to about $83 \pm 21$		559
XXXIII	Cancer	$142 \pm 54$	$3.11 \pm 0.50$	78 ± 22	4.37 ± 0.41	8
XXXIV	Cancer	$151 \pm 53$	$3.08 \pm 0.51$	78 ± 21	$4.31 \pm 0.42$	7
XXXV	Cancer	$219 \pm 62$		$73 \pm 15$		397
XXXVI	Crohn's disease	$165 \pm 100$	$3.0 \pm 1.5$	$66 \pm 30$	$4.1 \pm 0.8$	409
XXXVII	Gliomas (brain tu- mor)	134 ± 48		74 ± 18		462
XXXVIII	Cirrhosis	$71 \pm 43$		$70.3 \pm 2.54$		114
XXXIX	Cirrhosis	Decrease to about 50		About 90		564
XL	Cirrhosis	$19.5 \pm 7.5$		$67.3 \pm 18.5$		36
XLI	Cirrhosis	$63 \pm 30$	$2.55 \pm 1.5$	$66 \pm 30$	$4.1 \pm 0.8$	409
XLII	Liver disease	$17 \pm 3$	$3.5 \pm 1$	$30 \pm 20$	$4.5 \pm 0.5$	187
XLIII	Liver disease	$20 \pm 10$	<b>N</b>	$25 \pm 5$		185
XLIV	Chronic ulcerative colitis	Increase to 387 ± 68.7, depending on sever- ity	Decrease to 2.02 ± 0.5 depend- ing on severity	$80.7 \pm 14$	3.7 ± 0.3	137
XLV	Psychiatric pa- tients	$107 \pm 26$	4.4 ± 3.7	81 ± 22	$4.6 \pm 2.3$	86
XLVI	Depressed pa- tients	94 ± 30		62 ± 21		101
XLVII	Epil <del>e</del> psy disease	75 ± 10 after carbamazepine treat- ment, but 55 ± 10 after carbamazepine + phenobarbital treatment		75 ± 10		<del>89</del> –92
XLVIII	Epilepsy disease	$75 \pm 40$ during carbamazepine treat- ment, $100 \pm 50$ dur- ing phenytoin treat- ment		60 ± 35		531
IL	Epilepsy disease	$104.8 \pm 50$		$63.8 \pm 20$		441
L	Elderly patients with acute dis- ease	165 ± 59		63 ± 12		398
LI	Sick elderly pa- tients	$150 \pm 56$		62 ± 11 (young) 73 ± 15 (elderly)		397
LII	Genetic and envi- ronmental fac- tors			63 ± 18 (fathers) 62 ± 18 (mothers) 62 ± 18 (children)	$\begin{array}{l} 4.43 \pm 0.41 \\ 4.34 \pm 0.43 \\ 4.71 \pm 0.48 \end{array}$	16
LIII	Chronic hemodi- alysis patients	Before hemodialysis, $117.1 \pm 37.5$ ; after hemodialysis, 148.8 $\pm$ 44.7		65.6 ± 13.1		589
LIV	Hemodialysis	$60 \pm 15$		$25 \pm 5$		185
LV	Hepatitis	$62 \pm 25$		$70.3 \pm 2.54$		114
LVI	Hepatitis	About 90		About 90		564
LVII	Hypoalbuminemia	86 ± 29		$63 \pm 17$		436

PHARM REV



13	

		Disease le	evel	Normal k	evel	
No.	Disease	AGP [mg/100 ml]	HSA [g/100 ml]	AGP [mg/100 ml]	HSA [g/100 ml]	Ref.
LVIII	Infection	143 ± 64		73 ± 15		397
LIX	Inflammation of lungs	205 ± 38		76 ± 5		9
LX	Nonspecific upper respiratory tract infection	Increase to $140 \pm 12$		62 ± 28		426
LXI	Chronic obstruc- tive respiratory disease	139 ± 40		73 ± 15		397
LXII	Bacterial infection	Increase to about 280 ± 30		80 ± 30		553
LXIII	Viral infection	Increase to about 260 ± 30		80 ± 30		553
LXIV	Malaria	Increase to about 230 ± 30		80 ± 30		553
LXV	Systemic lupus er- ythematosis	180 ± 90	$3.6 \pm 0.8$	103 ± 35	$4.4 \pm 0.7$	139
LXVI	Lupus erythema- tosis	Increase to about 300		About 90		564
LXVII	Myocardial infarc- tion	Increase to $175 \pm 15$ at day 4	Decrease 45 0 45	79.6 ± 15.11		10
LXVIII	Myocardial infarc- tion	Increase to $99 \pm 5$ on day 3	Decrease to $3.45 \pm 0.11$ on day $3$	69 ± 4	3.7 ± 0.08	32
LXIX	Myocardial infarc- tion	Increase to $166 \pm 40$ on day 3		92 ± 20		34
LXX LXXI	Myocardial infarc- tion Myocardial infarc-	Increase to $160 \pm 20$ on about day 5 $05 \pm 27$ (males)		$100 \pm 10$		105
LXXII	Myocardial infarc- tion Myocardial infarc-	$95 \pm 27$ (males) $84 \pm 24$ (females)		$69 \pm 15 \text{ (males)}$ $72 \pm 15 \text{ (females)}$ $65 \pm 9$		106 169
LAAH	tion	181 $\pm$ 69; however, when chest pain, only 125 $\pm$ 37		00 X 9		109
LXXIII	Myocardial infarc- tion	Increase to $143 \pm 13$		$70 \pm 2.4$		504
LXXIV	Myocardial infarc- tion	Increase to about 175		80 ± 30		553
LXXV	Acute myocardial infarction	Increase to 170 on day 5	Decrease to 3 on day 5	93 ± 7	$3.85 \pm 0.1$	131
LXXVI	Acute myocardial infarction	$181 \pm 69$ ; however, when chest pain, only $125 \pm 37$		65 ± 9		169
LXXVII	Myocardial infarc- tion	$152 \pm 70$		73 ± 15		397
XXVIII	Obesity in men	$124 \pm 35$	$4.17 \pm 0.27$	$55 \pm 6$	$4.31 \pm 0.17$	47
LXXIX	Obesity	$121 \pm 17$	$4.0 \pm 0.2$	$62.9 \pm 18.8$	$4.2 \pm 0.2$	48
LXXX	Oral contracep- tives use	Decrease to about 56	Decrease to about 3.2	$71 \pm 17$	4.7 ± 0.47	196
LXXXI	Oral contracep- tives use	<b>54 ±</b> 15	3.9 ± 0.3	77 ± 30 (men) 64 ± 30 (women)	$4.3 \pm 0.6$ (men) $4.1 \pm 0.3$ (women)	408
LXXXII	Chronic pain	137 ± 8		81 ± 7		178
XXXIII	Acute pancreatitis	$220 \pm 48$		64 ± 16		427
XXXIV	Pregnancy	Decrease to $51 \pm 29$ in 3rd trimester; how- ever, when inflam- mation present, in- crease to about 200		62.6 ± 18.8		117
LXXXV	Pregnancy	85.1 ± 19.7	$3.1 \pm 0.3$	$107.1 \pm 19.4$	$4.3 \pm 0.3$	183
XXXVI	Pregnancy	Decrease to about $52 \pm 17$	Decrease to about $3.2 \pm 0.4$	71 ± 17	4.7 ± 0.47	196
XXXVII	Pregnancy	52 ± 9		<b>64 ± 16</b>		427



		Disease l	evel	Normal le		
No.	Disease					Ref.
		AGP [mg/100 ml]	HSA [g/100 ml]	AGP [mg/100 ml]	HSA [g/100 ml]	1001.
LXXXVIII	Pregnancy	In fetal serum, increase from 5 to 20 during gestation from week 15 to 40; in maternal	In fetal serum, increase from 1.5 to 3.5 dur- ing gestation	$25 \pm 10$ at birth	3.5 ± 0.5 at birth	284
		serum, 70 ± 30	from week 15 to 40; in mater- nal serum, $3 \pm$ 0.5	70 ± 30	3 ± 0.5	
LXXXIX	Pregnancy	49.7 ± 13.0 (maternal) 20.1 ± 94 (fetal)	3.56 ± 0.36 (ma- ternal) 4.35 ± 0.40 (fetal)			367
xc	Pregnancy	$15.3 \pm 4.7$ (fetal) 49.6 ± 6.5 (maternal)				585
XCI	Pregnancy	$72.1 \pm 2.7$ (maternal) $31.6 \pm 2.0$ (fetal)				300
XCII	Pregnancy	$54.5 \pm 3.7$	$2.48 \pm 0.76$	66.6 ± 18.2	$4.64 \pm 0.71$	140
XCIII	Pregnancy	In fetal serum, increase from $10 \pm 5$ to $30 \pm$ 10 during gestation from week 28 to 40		$30 \pm 25$ at birth and about $50 \pm 25$ on day 4		70
XCIV	Renal disease	$135 \pm 50$	$3.3 \pm 1.0$	$66 \pm 30$	$4.0 \pm 1.0$	206
XCV	Renal disease	$165 \pm 100$ (compli- cated); $82 \pm 40$ (un- complicated)	$2.6 \pm 1.5$ (compli- cated); $3.5 \pm$ 1.5 (uncompli-	$66 \pm 20$	$4.1 \pm 0.8$	409
			cated)			
XCVI	Septicemia	$247 \pm 45$		$64 \pm 16$		427
XCVII	Smoking	$112 \pm 63$		$92 \pm 25$		241
XCVIII	Smoking	$84.3 \pm 12.7$	$4.05 \pm 0.29$	$62.8 \pm 13.3$	$4.30 \pm 0.23$	46
IC	Traumatic injury	$197 \pm 100$		$70 \pm 16$	$4.8 \pm 0.4$	153, 154
С	Trauma	Increase to 243 be- tween days 10 and 14		$70 \pm 16$		155
CI	Surgical trauma	Increase to $216 \pm 35.2$ on day 5	Decrease to $4.1 \pm 0.3$ on day 4	Preoperative, 111.5 $\pm$ 29.1	Preoperative, $4.8 \pm 0.4$	24
CII CIII	Surgical trauma Postoperative cholecystectomy	$200 \pm 50$ $121 \pm 22$		$\begin{array}{c} 115 \pm 25 \\ 64 \pm 16 \end{array}$		129 427
CIV	Hip replacement	Increase to about 180		$80 \pm 30$		553
CV	Hernia repair	Increase to about 166 ± 28 on day 6, but increase to about 500 ± 100 when pneu- monia as complica- tion was diagnosed postoperatively	Decrease to about 2.59 ± 0.47 on day 6	Preoperative, 86 ± 21	3.43 ± 0.53	564
CVI	Surgical trauma	Increase to about 180		Preoperative, about $80 \pm 40$		163
CVII	Uremic patients	$184 \pm 62$		Range, 62–142		226
CVIII	Chronic inactive pyelonephritis	Increase to $240 \pm 40$		$62 \pm 28$		426
CIX	Nephritis	Increase to about 300		About 90		564
CX	Uremic patients	$25 \pm 15$		$25 \pm 5$		185
CXI	Uremic patients on hemodialysis	Before dialysis, 117.58 ± 36.65; after di- alysis, 132.19 ± 37.12		98.32 ± 19.5		147
CXII	<b>Vas</b> culitis	$30 \pm 10$		$25 \pm 5$		185
						100

growth factor receptor. Board et al. (64) reported recently, however, using the molecular cloning technique, that the homology between AGP and the epidermal growth factor receptor was poor.

Gamberg and Andersson (180) reported the presence of a membrane-bound form of AGP (with an apparent molecular weight of 52,000) on normal human lymphocytes, granulocytes, and monocytes (17-19). They demonstrated that this membrane protein is synthesized by lymphocytes and subsequently cleaved and released in the soluble serum form that has the normal molecular weight of 41,000. They concluded that this finding may partially explain the increase in the AGP level in serum in many disorders involving leucocyte proliferations (17– 19, 256, 257, 469).

There is also evidence that levels of AGP may change in the plasma of patients and several animals after treatment with defined drugs due to enzyme induction or inhibition of the AGP production (29, 33, 39, 84, 89-92, 99, 138, 301, 330, 386, 424, 425, 429, 432, 441, 531). This phenomenon has been reviewed by Greim (203). Both an increase and a decrease in the AGP level during drug therapy have been observed. Some studies with the same drug report contradictory results. Feely et al. (159) did not find an increase in the AGP level, due to hepatic enzyme induction of the AGP by rifampicin, as observed by Routledge et al. (441) and Delcroix et al. (138). Whereas Tiula and Neuvonen (531) and Olsson et al. (386) observed an increase in the AGP level after treatment with phenobarbital or carbamazepine alone, Bruguerolle et al. (89-92) observed a decrease in the AGP level only after treatment with a combination of these two drugs. Riva et al. (432) found an increase in the AGP level after carbamazepine treatment, in accord with the results of Tiula and Neuvonen (531) and Olsson et al. (386). Riva et al. (432) were the first to use serum from epileptic children and concluded that a modification of the serum AGP due to epilepsy itself cannot yet be ruled out and may be an explanation for the discrepancy in the results. Barbosa et al. (33) reported that anabolic steroids can both decrease and increase the AGP level, depending on their structures. As can be seen in table 6, nos. LXXXIV-XCIII and LXXX-LXXXI, the AGP level is decreased during pregnancy and during the use of oral contraceptives, due to the effect of estrogens. Reuss et al. (429) reported on a model that can be used to predict the AGP level during perazine therapy. Benedek et al. (46) found that smoking also raises the AGP levels, possibly due to an alteration in the serum protein chemistry or due to the accumulation of endogenous or exogenous substances (i.e., basic compounds in the smoke itself).

Many of the discrepancies in the literature dealing with changed AGP levels after drug therapies appear to be due to attempts by investigators to draw a single conclusion from studies involving different species and different dosing regimens.

## B. Several Other Biological Activities of Alpha-1-acid Glycoprotein

Many other biological properties of AGP are discussed in the literature: its immunological response behavior during several pathological states (50, 57, 110–112, 115, 253, 262, 405, 589); its protective effect against neonatal sepsis (57, 405); its inhibition of platelet aggregation (20, 21, 35, 124, 280, 342, 375, 378, 503, 507, 514, 569, 571); its interaction with collagen (124, 168); its growth-promoting effect for Hela and H-6 cells (325, 326); its involvement in the  $T_3$ - $T_i$  antigen-specific pathway of T- cell activation (511); its interaction with phospholipid membranes (110–112, 324–326, 368–372); its occurrence as carrier of a cofactor in the lipoprotein lipase reaction (509); its inhibition of phagocytosis (388, 389); its inhibition of the multiplication of malaria parasites (174, 175, 208); its interaction with vitamin  $B_{12}$  (224); its inhibition of neutrophil activation (125); its prolongation of the survival of skin homografts (361, 388); and its histamine binding capacity (103).

Bennett and Schmid (50) reported that the effectiveness of immunosuppression is enhanced for agalacto/ asialo derivatives of AGP; this points to the importance of the carbohydrate moiety in the immunoregulatory function of AGP. Cheresh et al. (112) studied sera from cancer patients and found a positive correlation between the AGP level and its immunosuppressive capacity. They observed the inhibitory effect of breast cancer serum on mitogen-induced blastogenesis of normal lymphoid cells. Cheresh et al. (111) further reported that nonspecific immunosuppression is due to electrostatic forces between sialic acids groups of AGP and phospholipids; however, no change in the lipid packing is involved because the phase transition temperature did not change. Jamieson et al. (253) found that AGP is located at the inflammatory site and may be involved in some aspects of the inflammation process.

Andersen and Eika (21) reported that totally desialylated AGP lost much of its capacity as inhibitor of the platelet aggregation, whereas Costello et al. (124) observed an increase in this effect in the presence of desialylated AGP. Barclay et al. (35) and Spragg et al. (507) observed that the inhibition of the hemagglutination was dependent on the shape and the size of the AGP polymers.

Franzblau et al. (67) reported that the interaction of AGP with collagen resulted in the formation of fibrous, long spacing fibers of collagen; presumably this process is also involved in the wound healing.

Maeda et al. (325, 326) found that AGP facilitated the passage of erythrocytes through membranes. AGP increased the bilayer thickness of liposomes and decreased the membrane permeability for ions (368, 369). More recently Neitchev (370) reported that the decreased permeability of liposomes after the addition of AGP was dependent on the AGP/protein ratio and was due to the interaction of AGP and protein with lipids, which in turn led to electrostatic changes in the membrane lipid region and membrane surface. Furthermore AGP could play the role of active modifier changing the membrane selectivity (371, 372).

Friedman et al. (174, 175) found that AGP could inhibit invasion by malaria parasites. However, Gupta et al. (208) could not confirm these inhibitory effects of AGP.

Chachaj et al. (103) studied the histamine-binding properties of plasma proteins. Their results suggested that human serum contains three histamine-binding

fractions, identified as orosomucoid and two glycoproteins belonging to the  $\alpha_1$ -globulin group. Parrot et al. (393) and Laborde et al. (295) reported earlier that serum from patients with allergic disorders showed an impaired ability to bind histamine. The increase in the binding to histamine in allergic diseases might perhaps be ascribed to an increase in AGP, as has often been reported in other inflammatory diseases, but the possible existence of such a correlation has not been studied so far.

These many diverse activities are difficult to interpret. By careful study of the papers cited it may become apparent that many of the reported activities do not occur in a regulatory fashion at physiological concentrations of AGP, and they are therefore unlikely to be functional (177, 342, 451). In view of the purpose of this review, we will not discuss these activities in detail here.

## IV. Interactions of Drugs with Alpha-1-acid Glycoprotein

HSA, AGP, and lipoproteins (LIPO) are the most important plasma proteins responsible for the binding of drugs in plasma. HSA binds in particular acidic and neutral drugs, whereas AGP and LIPO bind mainly basic drugs (82, 83, 194, 195, 358, 377, 406-408, 412, 414, 415, 450, 546). Table 7 gives a survey of studies which deal with the binding of drugs to AGP. Most of these drugs are basic ones with pK values of 8 or higher, which implies that these drugs are positively charged at physiological pH. Some of the drugs, such as phenylbutazone, phenobarbital, and the anticoagulants, are acidic and may be partially or totally negatively charged at neutral pH. Some other drugs, such as the steroids, diazepam, and carbamazepine, are neutral. From more recent studies dealing with the binding of drugs to AGP, it follows that other drugs not included in table 7 have an affinity for AGP; e.g., aminopyrine (156), amoxapine (160), bupropion (160), maprotiline (160), nomifensine (160), trazodone (160), drugs with a quaternary ammonium group (498), ritodrine (204), doxazosin (160), trimazosin (160), binedalin (358, 359), amsacrine (399), apazone (544) and SKF 525 A (45).

In this section the binding of drugs to AGP will be discussed. Section IV A will deal mainly with the relation between the varying concentration of AGP and its drug binding properties. In section IV B, the binding of basic and neutral drugs to isolated AGP is reviewed. A small section (IV C) is devoted to the binding of acidic drugs to AGP. The molecular details of the drug binding to AGP will be discussed in section IV D. In section IV E, it will be demonstrated that the results of binding studies can be strongly influenced by the experimental circumstances.

## A. Binding of Drugs to Alpha-1-acid Glycoprotein in Vivo

Since the end of the sixties it has become clear that AGP can function as a drug carrier for steroids (181, 182,

565). Later it was demonstrated that AGP also has high binding affinity for several basic drugs (396, 406-408) and, as has been shown recently, for some acidic drugs as well (249, 544, 545).

Variations have been observed in the binding of basic drugs in plasma (5, 396, 406-408, 538). This has been shown to be due to variations in plasma protein concentration, particularly in several disease states. Changes in the plasma protein concentrations have been reported for drug binding plasma proteins, particularly for HSA, AGP, and LIPO (406-408, 445, 538, 546). From binding studies it follows that HSA accounts mainly for the binding of acidic and neutral drugs, whereas AGP and LIPO associate more readily with basic drugs (194, 377, 406-408, 412, 414, 415, 546). It has become clear that the drug binding capacity of AGP, especially for basic drugs, can be of the same order as or even higher than that of HSA. This implies that the large variation in the AGP level in plasma observed during several physiological and pathological conditions can have a profound effect on drug concentrations in the blood. This correlation between the extent of drug binding and the AGP concentration in plasma will be discussed in the first part of this section.

It can be concluded from the literature (194, 377, 406–408, 412, 414, 415) that HSA, LIPO, and AGP are the most important plasma proteins that play a role in plasma drug binding. This means that the total drug concentration in plasma ( $c^{plasma}$ ) can be given by equation 1:

$$c^{\text{plasma}} = c_{\text{free}} + c^{\text{AGP}}_{\text{bound}} + c^{\text{HSA}}_{\text{bound}} + c^{\text{LIPO}}_{\text{bound}}$$
 equation 1

where  $c_{\text{free}}$  is the free concentration of the drug in plasma, and  $c_{\text{bound}}^{\text{AGP}}$ ,  $c_{\text{bound}}^{\text{HSA}}$ , and  $c_{\text{bound}}^{\text{LIPO}}$  represent the concentrations bound to AGP, HSA, and LIPO, respectively. If the binding of drugs to proteins can be described by Scatchard plots (196a, 349, 457), equation 2 can be used for the calculation of the drug concentrations bound to several components ( $c_{\text{bound}}^i$ ) in plasma:

$$c_{\text{bound}}^{\text{plasma}} = \sum c_{\text{bound}}^{i} = \sum \frac{n_i P_i K_i c_{\text{free}}}{1 + K_i c_{\text{free}}}$$
 equation 2

where  $n_i$ ,  $P_i$ , and  $K_i$  are the number of binding sites, the plasma protein concentration, and the affinity constant of component *i* in plasma, respectively. Note that, in this review,  $K_i$  is an association constant not a dissociation constant.

The use of equation 2 has been criticized in the literature (95). In the discussion that follows, it will become clear that equation 2 will apply to situations in which the average number of occupied binding sites is much less than one. In that case, there will no longer be objections to the use of this equation. Equations 1 and 2 describe the system completely. If protein concentrations, the number of binding sites, and the binding constants are known, then  $c_{tree}$  can be calculated at a

## DRUG BINDING TO HUMAN AGP IN HEALTH AND DISEASE

## TABLE 7 Survey of studies dealing with the binding to AGP

No.	Category	Drug	Ref.
I	Alpha-blocker	Nicergoline	433
п	Alpha-blocker	Prazosin	93, 136, 160, 200, 447, 542
ш	Anesthetic/analgesic	Alfentanil	245, 348
īv	Analgesic	Fentanil	348
v	Analgesic/anesthetic	Ketamine	135
vi	Analgesic	Meperedine	63, 242, 367
VI		Methadone	
	Analgesic		7, 136, 436, 521
VIII	Analgesic	Phenylbutazon	545
IX	Anesthetic	Bupivacaine	128, 136, 140, 404, 410
х	Anesthetic	Etidocaine	136, 357, 410
XI	Anesthetic/antiarrhythmic	Lidocaine	34, 63, 71, 134, 136, 155, 159, 191, 197, 206, 250, 285, 410, 411, 439-444, 493, 585
XII	Anesthetic	Phencyclidine	31, 190, 390
XIII	Antiarrhythmic	Aprindine	22, 524
XIV	Antierrhythmic	Disopyramide	10a, 71, 78, 79, 131, 133, 136, 186, 222, 223, 233, 244, 264, 313-317, 346, 397, 412, 414
XV	Antiarrhythmic	Quinidine	87, 88, 136, 154, 170–172, 210,
	·	•	251, 376, 377, 412, 415
XVI	Antiarrhythmic	Verapamil	136, 187, 191, 269, 338, 588
XVII	Antibiotic	Erythromycin	
		Acenocoumarol	36, 142, 420
XVIII	Anticoagulant		545
XIX	Anticoagulant	Dipyridamole	157, 176, 177, 281, 375, 514
XX	Anticoagulant	PCR 2362, thienopyridin derivative	195
XXI	Anticoagulant	Ticlopidine	195
XXII	Anticoagulant	Warfarin	544, 545
XXIII	Antiepileptic	Carbamazepine	37, <b>304, 339, 340, 4</b> 32, 531
XXIV	Antiepileptic	Phenytoin	45, 411, 462, 549
XXV	Antiinflammatory agent	Naproxen	408
XXVI	Beta-blocker	Alprenolol	42, 43, 75, 136, 235, 236, 408
XXVII	Beta-blocker	Metoprolol	42, 43
XXVIII	Beta-blocker	Oxprenolol	•
			42-45, 271
XXIX	Beta-blocker	Pindolol and 8 related compounds	42, 43, 305
XXX	Beta-blocker Beta-blocker	Propranolol Timolol	8, 11, 16, 41–48, 63, 88, 97, 160, 161, 177, 194, 238, 247, 251, 274, 282, 342, 359, 363, 394, 397, 398, 400, 409, 411, 412, 421, 443, 449, 450, 452, 480, 488, 505, 549, 551, 557, 582, 583, 585
XXXI			42, 43
XXXII	Estrogen	Progesterone	128, 177, 181, 182, 272, 274, 294, 442, 506, 555, 565–568, 582
XXXIII	Estrogen	Cortexone	182, 272
		Cortisol	
		Testosterone	
		Estradiol	
XXXIV	Neuromuscular blocker	Metocurine	585
XXXV	Neuromuscular blocker	d-Tubocurarine	585
XXXVI	Psychotropic	Amitriptyline	83, 85, 86, 160, 412, 413, 415,
	,		463
XXXVII	Psychotropic	Chlorpromazine	158, 176, 282, 359, 363, 409, 463, 549, 551
KXXVIII	Psychotropic	Cyclazindol	520
XXXIX	Psychotropic	Desmethylimipramine	84, 254, 255, 551
XL	Psychotropic	Diazepam	5, 84, 134, 206, 282, 332, 442, 462, 585
<b>NT 1</b>	Psychotropic	Doxepin	160, 552
XLI		-	-
XLI	Psychotropic	Fluphenazine	282



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No.	Category	Drug	Ref.
XLIV	Psychotropic	Imipramine	6, 63, 75, 160, 169, 192, 193, 254, 255, 282, 287, 288, 332, 359, 363, 364, 408, 462, 463
XLV	Psychotropic	Loxapine	282
XLVI	Psychotropic	Mianserin	491
XLVII	Psychotropic	Nortriptyline	82, 83, 160, 412
XLVIII	Psychotropic	Norzimelidine	101
IL	Psychotropic	Perazine	82, 83, 86, 136, 429, 460-463
L	Psychotropic	Perphenazine	551
LI	Psychotropic	Phenobarbital	462
LII	Psychotropic	Phenothiazine derivatives	158, 460, 461, 463, 549-551
LIII	Psychotropic	Promazine	158
		Acepromazine Protipendyl	
LIV	<b>Psychotropic</b>	Thioridazine	26, 282, 380, 381
LV	Psychotropic	Thiothixene	282, 463
LVI	Psychotropic	Triazolam	290
LVII	Psychotropic	Trifluoperazine	158, 551
LVIII	Psychotropic	Zimelidine	101
LIX	Vitamin	Vitamin B <sub>12</sub>	224
LX	Fluorescent probe	DAPN, derivative of propranolol	4
	• • • • •	1,8-Anilinonaphtalene sulfonate	97, 176

given total plasma concentration. Changes in one of the components (either drug or protein) cause several equilibria to shift until a new equilibrium is reached. However, there is no simple relationship between protein concentration and drug concentration. For the in vivo situation to be discussed below, simplifications can be made which result in equations that are easier to handle. The assumption is that  $K_i c_{free}$  is much smaller than one. If this holds, then equation 2 can be reduced to:

$$\Sigma c_{\text{bound}} = \Sigma n_i P_i K_i c_{\text{free}} \qquad \text{equation 3}$$

Substitution of equation 3 into equation 1 and dividing by  $c_{\text{free}}$  yield:

$$c^{\text{plasma}}/c_{\text{tree}} = 1 + \Sigma n_i P_i K_i$$
 equation 4

Defining  $c_{\text{free}}/c^{\text{plasma}}$  as the free fraction F, and 1-F as the bound fraction B, then:

$$1/F = 1 + n_{AGP}P_{AGP}K_{AGP} + n_{HSA}P_{HSA}K_{HSA} + n_{LIPO}P_{LIPO}K_{LIPO}$$
 equation 5  
$$R/F = n_{AGP}P_{AGP}K_{AGP} + n_{HSA}P_{HSA}K_{HSA}$$

$$+ n_{\text{LIPO}} P_{\text{LIPO}} K_{\text{LIPO}} \text{ equation 6}$$

Equations 5 and 6, which are much easier to deal with than equations 1 and 2, can be used to study the influence of varying AGP concentrations on B/F or 1/F. If the concentration of HSA and LIPO remains constant, then the binding ratio B/F varies linearly with the AGP concentration.

From the literature, data were collected on drug levels in plasma as a function of AGP concentration. Some authors had already analyzed their data according to equation 6. Others presented an analysis in which F was plotted versus the AGP concentration. However, as shown in equation 5, only for 1/F versus the AGP concentration could a linear relationship be expected. These relationships were replotted by us in the form of B/F and are denoted as "transformed" in table 8.

Table 8 presents a survey of those studies in which drug binding was measured as a function of AGP concentration. The drugs studied are listed in the second column. Except for prednisolone and triazolam, which are neutral, all the drugs are basic ones, and nearly all of them are positively charged at neutral pH. The next column gives the AGP range studied. The fourth column shows that the free fraction F might indeed vary strongly, and it is this variation which might have therapeutic consequences. Column 6 gives the linear correlation found between B/F(y) and the AGP concentration (x), expressed in mg/100 ml, together with the correlation coefficient r. As judged by this value of r, many good linear relationships are found, which means that the variation in F, and therefore in  $c_{\text{free}}$ , is caused by the variation in the AGP concentration. It should be noted that not all relationships reported in table 8 are significant, particularly when r becomes smaller than 0.7. Then  $r^2$  is smaller than 0.5, which in turn means that only 50% of the total variance can be explained by the observed relationship. The variation occurring in the HSA level in disease states (table 6) may be a possible reason.

The value of  $n_{AGP}K_{AGP}$  has been calculated from the slope of the relationships given in table 8 (column 7). Two compounds (VI and XXXII) show a relatively high value of  $n_{AGP}K_{AGP}$ . About 90% of the remaining values are found in a relatively small range between  $8 \times 10^4$  M<sup>-1</sup> and  $8 \times 10^5$  M<sup>-1</sup>. This is surprising and probably points to some common structural elements in these drugs. In addition, these values allowed us to verify that, for the most cases studied,  $Kc_{free}$  was indeed much smaller than one. Lidocaine might be an exception (285).

PHARMACOLOGICAL REVIEW

Survey of the studies dealing with the correlation between the binding ratio and the concentration of AGP in plasma/serum for several drugs and in several disease states

		ü	n several disea	se states			
No.	Drug	Disease	AGP range [mg/100 ml]	F range	B/F	$n_{AGP}K_{AGP} \times 10^{-6}$ [M <sup>-1</sup> ]	Ref.
I	Alprenolol	Healthy people	38-113	0.22-0.09	y = 0.053x + 1.97 r = 0.72	2.14	408, fig. 1, transformed
п	Alprenolol	Pregnancy (mother/ newborn)	8–70	0.44-0.12	y = 0.073x + 0.59 r = 0.72	2.95	236, fig. 7
ш	Amitriptyline	Depressive patients	17-118 (concentra- tion of S- variant of AGP)	0.12–0.02	y = 0.214x + 6.60 r = 0.73	8.56	529, fig. 3, transformed
***		Correlation with S-variant	•		-	0.05	
IV	Carbamazepine	Epilepsy	57-210	0.32-0.17	y = 0.006x + 3.15 r = 0.42	0.25	121, fig. 1, transformed
V	Carbamazepine- 10,11-epoxide	Epilepsy	60-210	0.51–0.33	y = 0.004x + 1.12 r = 0.54	0.16	121, fig. 1, transformed
VI	Chlorpromazine	Healthy people, renal failure, arthritis, Crohn's disease, cirrhoeis	40-258	0.022–0.005	y = 0.430x + 3.0 r = 0.82	17.2	409, fig. 2, transformed
VII	Disopyramide	Kidney transplant, recipients	35–190	0.68–0.05	y = 0.079x - 3.64 r = 0.85 (curvilin- ear!!)	3.16	222, fig. 1
VIII	Disopyramide	Acute myocardial in- farction	52-240	0.480.08	y = 0.027x + 0.04 r = 0.64	1.06	131, fig. 5, transformed
IX	Disopyramide	Patients on antiar- rhythmic therapy	48-300	0.58-0.15	y = 0.019x + 0.04 r = 0.96	0.76	78, fig. 3
х	Erythromycin	Cirrhosis	7–95	0.80-0.15	y = 0.060x + 0.13 r = 0.94	2.39	36, fig. 1, transformed
XI	Imipramine	Healthy people	36–114	0.11-0.06	y = 0.063x + 7.60 r = 0.79	2.53	408, fig. 2, transformed
XII	Imipramine	Age of healthy people	35–150	0.19-0.09	y = 0.015x + 5.31 r = 0.31	0.60	6, fig. 3, transformed
XIII	Imipramine	Severe burn injury	34-268	0.19–0.05	y = 0.044x + 3.63 r = 0.78	1.76	332, fig. 3, transformed
XIV	Imipramine	Healthy people, smoking, habits, sex, age, oral con- traceptives use, pregnant women	38–200	0.15–0.08	y = 0.034x + 4.81 r = 0.74	1.36	287, fig. 2, transformed
xv	Imipramine	Healthy people	47-160(?)	0.10-0.04	y = 0.076x + 5.48 r = 0.76	3.03	255, fig. 2, transformed
XVI	Imipramine	Cardiac patients	56-346	0.15-0.05	y = 0.026x + 7.88 r = 0.53	1.04	169, fig. 2
XVII	Imipramine	Isolated AGP (!!)	0–250	0-0.09	y = 0.041x - 0.19 r = 0.96	1.65	169, fig. 3
XVIII	Imipramine	Rheumatoid arthritis	75–270	0.15-0.07	y = 0.035x + 4.20 r = 0.82	1.40	288, fig. 1
XIX	Lidocaine	Pregnancy (mother/ newborn), oral contraceptives use	4–69	0.74-0.24	y = 0.026x + 0.57 r = 0.62	1.02	585, fig. 2
XX	Lidocaine	Epilepsy	45-165	0.39–0.17	y = 0.022x + 0.51 r = 0.91	0.87	441, fig. 2
XXI	Lidocaine	Healthy people	43–137	0.40-0.20	y = 0.026x + 0.50 r = 0.96	1.03	439, fig. 2
XXII	Lidocaine	Myocardial infarc- tion	58-152	0.38-0.21	y = 0.023x + 0.03 r = 0.89	0.92	440, fig. 1
XXIII	Lidocaine	Cancer	30–300	0.36-0.09	y = 0.031x + 0.54 r = 0.81	1.24	250, fig. 1, transformed
XXIV	Lidocaine	Renal disease	16–188	0.50-0.14	y = 0.027x + 0.34 r = 0.93	1.09	206, fig. 4
XXV	Lidocaine	Trauma	58-300	0.45-0.12	y = 0.020x + 0.16 r = 0.97	0.81	155, fig. 2

PHARM REV

PHARMACOLOGICAL REVIEWS

No.	Drug	Disease	AGP range [mg/100 ml]	F range	B/F	$\begin{array}{c} n_{AGP}K_{AGP} \\ \times 10^{-6} \\ [M^{-1}] \end{array}$	Ref.
XXVI	Lidocaine	Trauma, in four pa- tients observed	194-298	0.22-0.12	y = 0.029x - 1.83 r = 0.92	1.15	155, fig. 4
			112-230	0.31-0.18	y = 0.017x + 0.27 r = 0.79	0.69	
			136-244	0.31-0.15	y = 0.027x - 1.17 r = 0.96	1.08	
			120-195	0.31–0.16	y = 0.040x - 2.88 r = 0.99	1.62	
XXVII	Methadone	Healthy people	28-120	0.13-0.07	y = 0.054x + 5.21 r = 0.68	2.18	436, fig. 1b, transformed
ххуш	<b>Methad</b> one	Arthritis, hypoalbu- minemia, healthy people	25–160	0.21-0.04	y = 0.063x + 4.13 r = 0.70	2.52	436, fig. 2, transformed
XXIX	Methadone	Cancer	41-251	0.25-0.06	y = 0.035x + 1.25 r = 0.76	1.39	7, fig. 1
XXX	Meperedine	Pregnancy (mater- nal/fetal plasma)	13.3-68.8	0.53-0.31	y = 0.015x + 0.91 r = 0.60	0.60	367, fig. 1
XXXI	Nortriptyline	Depressive patients	35-148	0.14-0.02	y = 0.169x - 2.60 r = 0.49	6.78	529, fig. 2, transformed
ХХХП	No <del>rtript</del> yline	Depressive patients	19-118 (concentra- tion of S- variant of	0.12-0.02	y = 0.316x - 1.77 r = 0.77	12.6	530, fig. 4, transformed
			AGP) ant of AGP more	evident than w	ith total plasma AGP		
XXXIII	Perazine	Schizophrenic pa- tienta	45-124	0.044-0.029	y = 0.091x + 19.1 r = 0.63	3.65	465, fig. 3, transformed
XXXIV	Perazine	Psychiatric patients	67-163	0.06-0.03	y = 0.122x + 9.08 r = 0.79	4.87	84, fig. 2, transformed
XXXV	Perazine	Healthy people	53-140	0.06-0.03	y = 0.086x + 14.8 r = 0.38	3.43	84, fig. 2, transformed
XXXVI	Prazosin	Healthy people	64-284(?)	0.10-0.025	y = 0.149x + 2.55 r = 0.97	5.74	130, fig. 4B
XXXVII	Prednisolone (neutral drug!)	Isolated AGP (!!)	25500	0.98-0.58	y = 0.001x + 0.074 r = 0.91	0.04	353, fig. 7
XXXVIII	Propranolol	Pregnancy (mother/ newborn) oral con- traceptives used	4-64	0.45-0.12	y = 0.069x + 2.03 r = 0.66	2.77	585, fig. 3
XXXIX	Propranolol	Elderly patients with acute illness	44-313	0.25-0.02	y = 0.118x - 0.91 r = 0.85	4.6	397, fig. 1
XL	Propranolol	Acute myocardial in- farction	24-184	0.14-0.04	y = 0.072x + 7.39 r = 0.66	2.89	449, fig. 1
XLI	Propranolol	Healthy people	48-204(?)	0.12-0.05	y = 0.064x + 5.40 r = 0.85	2.56	450, fig. 4
XLII	Propranolol	Healthy people, renal failure, arthritis, Crohn's disease, cirrhosis	40-268	0.18-0.03	y = 0.071x + 3.80 r = 0.77	2.85	409, fig. 1, transformed
XLIII	Propranolol	Healthy people, el- derly patients with acute illness	45-295	0.22-0.03	y = 0.095x - 0.02 r = 0.88	3.81	396, fig. 1
XLIV	Propranolol	Healthy people, el- derly patients with acute illness	<b>46–27</b> 2	0.22-0.03	y = 0.096x + 0.02 r = 0.88	3.85	398, fig. 3
XLV	Propranolol	Smoking effect	43-100	0.20-0.0 <del>9</del>	y = 0.081x + 0.92 r = 0.73	<b>3.26</b>	46, fig. 1
XLVI	Propranolol	Obesity	<b>36</b> 133	0.15-0.07	y = 0.058x + 3.67 r = 0.88	2.32	<b>48, fig.</b> 1
XLVII	Propranolol	Cancer and its treat- ments	41-256	0.280.06	y = 0.053x + 0.45 r = 0.93	2.1	8, fig. 1
XLVIII	Propranolol	Healthy people	<b>23</b> –132	0.14-0.02	y = 0.175x + 5.50 r = 0.67	7	16, fig. 3



No.	Drug	Discase	AGP range [mg/100 ml]	F range	B/F	$n_{AGP}K_{AGP} \times 10^{-6}$ [M <sup>-1</sup> ]	Ref.	
IL	Propranolol	Moderately obese male subjects	45–190	0.15-0.08	y = 0.030x + 4.43 r = 0.73	1.20	47, fig. 1	
L	Propranolol	Arthritis, dissemi- nated lupus, can- cer, bacterial infec- tion	90-410	0.11-0.02	y = 0.135x + 0.63 r = 0.88	5.4	107, fig. 4	
LI	Propranolol	Isolated AGP (!!)	20-400	0.50-0.07	y = 0.028x - 0.05 r = 0.97	1.13	107, fig. 1	
LII	Propranolol	Isolated AGP in presence of HSA (4 g/100 ml) (!!)	20-400	0.17-0.02	y = 0.096x + 3.59 r = 1.0	3.84	107, fig. 1	

TABLE 8—Continued

 $\rightarrow$  isolated AGP does not behave like AGP in serum; HSA potentiates binding to AGP; extent of potentation depends on lipids associated with

			AGP ana n	ISA			
LIII	Propranolol	Malnutrition	45-200	0.30-0.07	y = 0.039x + 1.60 r = 0.71	1.56	252, fig. 1, transformed
LIV	Quinidine	Healthy people	48-208(?)	0.33-0.15	y = 0.017x + 1.71 r = 0.86	0.67	377, fig. 3
LV	Quinidine	Traumatic injury	113-300	0.11-0.05	y = 0.051x + 2.94 r = 0.88	2.04	154, fig. 3
LVI	Triazolam	Patients on dialysis	72–205	0.15-0.06	y = 0.051x + 3.21 r = 0.82	2.05	290, fig. 1
LVII	Verapamil	Liver disease	14-58	0.23-0.07	y = 0.174x + 3.72 r = 0.80	<b>7.09</b>	187, fig. 3b
LVIII	Verapamil	Healthy people	40-130	0.11-0.06	y = 0.046x + 6.85 r = 0.79	1.85	338, fig. 1
LIX	Verapamil	Isolated AGP (!!)	0–150	0.44-0.06	y = 0.082x + 0.95 r = 0.99	3.29	338, fig. 2
LX	Zimelidine	Depressed patients	38-105	0.13-0.05	y = 0.113x + 3.11 r = 0.73	4.5	101, fig. 3, transformed
LXI	Norzimelidine	Depressed patients	3 <b>9</b> –107	0.32-0.23	y = 0.012x + 1.75 r = 0.69	0.46	101, fig. 3, transformed

\* F, free fraction; B, bound fraction; B/F, binding ratio (under this heading the relation is given between B/F and the AGP concentration, as expressed by equation 6); y, B/F; x, AGP concentration in mg/100 ml; r, correlation coefficient;  $n_{AGP}$ , number of binding sites on AGP;  $K_{AGP}$ , association constant of drug for AGP; the value of  $n_{AGP}K_{AGP}$  is obtained from equation 6 and given in M<sup>-1</sup>, assuming a molecular weight of 40,000.

† !!, study done with isolated AGP (in vitro) instead of plasma/serum (in vivo); ?, very high value reported for AGP in serum of normals.

It is not possible to discuss all relationships in table 8 in detail. Instead, in table 9 the data relating to propranolol have been given. Propranolol was chosen because many studies have been done on this compound. Only those studies having  $r^2 \ge 0.5$  were taken from table 8. The F test of significance indicates a higher than 99% probability for these relationships. If one takes into account that these data originate from different sources and refer to different diseases, the correspondence in the slope of the various linear relationships is remarkable. According to equation 6, this slope represents  $n_{AGP}K_{AGP}$ , which should be constant as long as the molecular properties of the AGP have not changed. So despite the fact that heterogeneity can be expected due to the diseases (see section II D), it seems that this heterogeneity does not strongly influence the binding constants of propranolol.

The value of the constant term in the linear relationships in this column of table 9 represents the contribution of HSA and LIPO to B/F. This varies in the various cases described. According to equation 6, this is due to variation in the concentration and number of binding sites of HSA and LIPO.

The linear relationships also allow us to calculate the contribution of AGP to the fraction bound. Examples are given in table 9. By substituting a given AGP concentration in the linear relationship (second column), the corresponding B/F, and therefore B value, can easily be calculated. This has been done for three values of the AGP concentration, namely, zero, 73, and 219 mg/100 ml. Note that the average value of the AGP concentration in normal conditions (see section III A) is 73 mg/100 ml. In comparing these B values, one should take the standard errors into account. It is then evident that the calculated values of B as predicted from the various studies are very consistent, as can be seen from the last three columns in table 9.

The linear relationships further permit one to calculate the contribution of AGP to the total binding in the following way. The value of  $(B/F)_{AGP}$  can be calculated for a given value of the AGP concentration. Dividing this by the value of the total B/F gives a number which

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represents the fraction of the total drug bound that is accounted for by AGP. The last column in table 9 gives some numbers for an AGP concentration of 73 mg/100 ml. The results of the various studies show that AGP accounts for more than 50% of the binding.

In table 10 the binding parameters in plasma for AGP and HSA are summarized. These data were obtained from the fitting of Scatchard plots measured in plasma on the assumption of the presence on AGP and HSA of two classes of binding sites, one with a high affinity and a low capacity and the other one with a low affinity and a high capacity. From these parameters too it is possible to estimate the relative contribution of AGP and HSA to the total plasma binding. This method is a different way of approaching the problem discussed in this section. However, as only a very limited number of studies have been reported, a detailed comparison cannot be made. Pike et al. (412, 414, 415) and Suzuki et al. (517) used plasma that was deficient in several plasma protein fractions in order to study the binding of acidic, neutral, and basic drugs in plasma. They found that the binding of basic drugs decreased considerably in AGP-deficient plasma. A decrease was observed for acidic and neutral drugs only in HSA-deficient plasma in accordance with the evidence presented above.

It is clear that the data presented in this section confirm that AGP makes an important contribution to the binding of many drugs.

Displacement studies have been performed in order to obtain information about the possibility of clinically relevant competition phenomena in vivo (158, 197, 337, 343, 362–364, 401, 460–465, 544, 581, 588). McElnay and D'Arcy (337) reported recently that the clinical importance of drug displacement during combined drug ther-

 TABLE 9
 Further analysis of propranolol binding data from table 8

				ι,			
No.*	$y = ax + b\dagger$	F test of significance	r <sup>2</sup>	<i>B</i> [AGP] = 0	<i>B</i> [AGP] = 73 mg/100 ml	B [AGP] = 219 mg/100 ml	Fraction of bound drug, bound by AGP at [AGP] = 73 mg/100 ml
XXXIX	$y = 0.118 \ (0.007) \ x - 0.91 \ (0.86)$	1,102 = 285	0.74		0.88 (0.01)	0.96 (0.003)	1.12 (0.16)
XLI	$y = 0.064 \ (0.009) x + 5.40 \ (1.10)$	1,19 = 48	0.72	0.84 (0.03)	0.91 (0.01)	0.95 (0.005)	0.46 (0.09)
XLII	$y = 0.071 \ (0.007) \ x + 3.80 \ (0.80)$	1,78 = 111	0.59	0.79 (0.03)	0.90 (0.01)	0.95 (0.004)	0.58 (0.08)
XLIII	y = 0.095 (0.007)x - 0.01 (0.086)	1,55 = 185	0.77	0.00 (1.06)	0.87 (0.02)	0.95 (0.004)	1.00 (0.16)
XLIV	$y = 0.096 \ (0.008) \ x - 0.02 \ (0.81)$	1,40 = 138	0.78	0.00 (1.27)	0.87 (0.02)	0.95 (0.004)	1.00 (0.16)
XLVI	y = 0.058 (0.010)x + 3.67 (0.93)	1,9 = 31	0.78	0.79 (0.04)	0.89 (0.01)	0.94 (0.008)	0.54 (0.12)
XLVII	$y = 0.053 \ (0.004) x + 0.45 \ (0.56)$	1,21 = 137	0.87	0.31 (0.27)	0.81 (0.02)	0.92 (0.006)	0.90 (0.15)
L	$y = 0.135 \ (0.019) x + 0.63 \ (3.39)$	1,14 = 51	0.79	0.39 (1.28)	0.91 (0.03)	0.97 (0.006)	0.94 (0.35)
LI	$y = 0.028 \ (0.002) x + 0.05 \ (0.41)$	1,12 = 216	0.95	0.05 (0.37)	0.68 (0.05)	0.86 (0.012)	0.98 (0.21)
LII	$y = 0.096 \ (0.002) \ x + 3.59 \ (0.42)$	1,11 = 2,224	0.99	0.78 (0.02)	0.91 (0.00)	0.96 (0.001)	0.66 (0.03)
LIII	y = 0.039 (0.006)x + 1.60 (0.68)	1,38 = 38	0.50	0.61 (0.10)	0.82 (0.03)	0.91 (0.012)	0.64 (0.15)

\* The numbers in the first column refer to the compounds in table 8.

† In the second column, y represents B/F and x represents the AGP concentration in mg/100 ml. Numbers in parentheses in this and in other columns in this table represent the standard error in this parameter. The standard error in the value of B, denoted by  $S_B$ , follows from the relationship  $S_B = (1 + y)^{-2} \cdot S_y$ .

		TABL	Е	10				
Survey of	f binding	parameters	for	AGP	and	HSA	in	pla

No.	Category	Drug	nagpPagp*	$K_{AGP} [M^{-1}]^{\dagger}$	n <sub>HSA</sub> P <sub>HSA</sub>	К <sub>нза</sub> [м <sup>-1</sup> ]	Ref.
I	Anesthetic	Bupivacaine	рН 7.4 -		pH	140	
			pH depe		•	undependent	
			$(1.56 \rightarrow 2.14) \times 10^{-5}$	$(1.69 \rightarrow 0.6) \times$	$0.53 \rightarrow 0.21$	$(4.21 \rightarrow 5.03) \times 10^3$	
				10 <sup>6</sup>			
п	Tricyclic antidepressant	Amitriptyline	$1.6 \times 10^{-4}$	$5.9  imes 10^{4}$	$4.1  imes 10^{-3}$	$7.3  imes 10^{2}$	83
ш	Tricyclic antidepressant	Nortriptyline	$3.6 \times 10^{-4}$	$1.8 \times 10^{4}$	$1.8 \times 10^{-3}$	$1.4 \times 10^{3}$	83
· IV	Tranquilizer	Thioridazine		$6.39 \times 10^{7}$			381
v	Beta-blocker	Alprenolol		$(3-5) \times 10^{5}$			236
VI	Beta-blocker	Oxprenolol	8 × 10 <sup>−6</sup>	$1.3 \times 10^{6}$			44
VII	Beta-blocker	Propranolol	Binding in serum defi	cient in AGP decre	ases, but no effe	ct in serum deficient	412
				in H	SA		414
VIII	Beta-blocker	Propranolol	Stereoselective bindin	g to AGP of same of	order as to plasm	a, but different in	557
		•		HS	A		
IX	Beta-blocker	Propranolol	$2.04 \times 10^{-5}$	$5.87 \times 10^{5}$			44
х	<b>Antiarr</b> hythmic	Quinidine	Binding in a	erum deficient in A	AGP decreases c	onsiderably	412
	-	Quinidine	$3.49 \times 10^{-5}$	$1.17 \times 10^{5}$	$3.14 \times 10^{-3}$	$1.33 \times 10^{3}$	154
XI	Antiepileptic	Carbamazepine	2.2	$2.4  imes 10^{4}$	9	$4.6  imes 10^2$	340

\* n<sub>AGP</sub>P<sub>AGP</sub> and n<sub>HSA</sub>P<sub>HSA</sub>, binding capacity to AGP and HSA, respectively, in plasma, using Scatchard plots.

 $\dagger K_{AGP}$  and  $K_{HBA}$ , affinity constant to AGP and HSA, respectively, in plasma, calculated from Scatchard plots.

apy has been overestimated because physiological drug concentrations are generally lower than the concentrations used in in vitro studies. Goolkasian et al. (197), who studied the displacement of lidocaine, concluded from their results that a clinically significant displacement interaction of the drugs studied occurs only when bupivacaine and lidocaine are used in combination. McNamara et al. (343) reported earlier that clinical concentrations of bupivacaine, disopyramide, and quinidine increase the lidocaine concentrations. Mueller et al. (364), who studied the drug displacement between psychotropics, concluded that competition phenomena in vivo may occur for methaqualone and thioridazine. Further binding studies in plasma using several combinations of drug will probably give more information about the clinical relevance of these competition phenomena in vivo.

## B. Binding of Basic and Neutral Drugs to Alpha-1-acid Glycoprotein in Vitro

The binding of drugs to AGP in plasma has been discussed in section IV A. A different approach to the study of drug-protein interaction can be followed by first isolating the binding protein from plasma, redissolving the isolated protein in an appropriate solvent (generally an aqueous buffer solution), and using this protein solution for binding experiments. Studies of this type performed with AGP or HSA will be referred to as isolated AGP or isolated HSA binding studies, in order to distinguish them from the binding studies in plasma. Sometimes a mixture of AGP and HSA was used. Results of studies of this type done on isolated AGP or HSA are collected in table 11. In columns 4 and 5 of table 11, the free fractions measured in solutions of isolated AGP or HSA at variable concentrations ( $F_{AGP}$  and  $F_{HSA}$ , respectively) are reported so that the contribution that each of these proteins makes to the total plasma binding can be estimated. In columns 6 to 9 of table 11, the number of binding sites  $(n_{AGP} \text{ and } n_{HSA})$  and the binding constants  $(K_{AGP} \text{ and } K_{HSA})$  of drugs for isolated AGP and HSA are given.

One of the main purposes of this table is to collect data scattered throughout the literature. For a given drug, the combined in vivo and in vitro data can give a picture of the importance of protein binding. Because of the vast amount of data available, it is not possible to discuss the various compounds. Only some general comments will be made.

In a comparison of the binding parameters of the same drug obtained in different studies, it should be noted that the AGP samples were obtained using different methods and that these can have different effects on the physicalchemical properties of AGP, as discussed in section II. It should also be pointed out that, since the AGP concentrations used to determine the free fraction were not the same, different values for the free fraction may result. From studies on binding in solutions of AGP, it follows that often two classes of binding sites are present on AGP. Therefore the in vitro results cannot be compared indiscriminately with the results obtained in plasma reported in section IV A.

Other factors influencing the binding parameters are discussed below in section IV E.

## C. Binding of Acidic Drugs to Alpha-1-acid Glycoprotein in Vitro

It is generally assumed that in plasma acidic drugs are mainly bound to HSA. Four recent studies (249, 462, 544, 545) have shown, however, that the association constants of some acidic drugs to AGP are high enough to indicate that binding to AGP will contribute significantly to the total plasma binding of these drugs, especially in diseases in which the concentration of AGP increases and/or of HSA decreases.

The parameters describing the binding of acidic drugs to isolated AGP are summarized in table 12. Israili and El-Attar (249) found that the binding to AGP increased with increasing concentration of AGP and decreasing concentration of the drugs (therapeutic range). The maximum binding of each drug to AGP (at 200 mg/100 ml) was, however, always lower than the binding to HSA (at 4.5 g/100 ml).

Urien et al. (545) studied several acidic drugs with or without a carboxylic group and found that clofibric, fenofibric, salicylic, and valproic acid do not bind to AGP, and that benoxaprofen, indomethacin, and itanoxone at a molar drug/AGP ratio of 0.04 (AGP concentration, 90 mg/100 ml) bind very poorly (table 12). In contrast, the percentages of bound warfarin, acenocoumarol, and phenylbutazone are noticeably higher. The acidic drugs which exhibit a high or intermediate affinity to AGP do not exhibit any carboxyl moiety and share a common specific binding site on HSA, called site I or the warfarin site. By contrast, all the drugs having a poor affinity or no affinity to AGP exhibit carboxyl groups and bind specifically to another HSA binding site, called site II or the diazepam site (527). Moreover, these results demonstrate the existence of only one binding site on AGP, which is the result found earlier for basic drugs. For these acidic drugs, Urien et al. (545) made some calculations to estimate the relative contribution of the drug bound to HSA and AGP in plasma. Acenocoumarol. phenylbutazone, and warfarin would then bind for 90%, 99.3%, and 94%, respectively, to HSA and for 9%, 0.3%, and 5%, respectively, to AGP, whereas the sum of the binding to HSA and AGP is close to the value observed in human plasma.

Comparing the results of Urien et al. (545) and Israili and El-Attar (249), one can conclude that Urien et al. found a much higher affinity of phenylbutazone for AGP than Israili and El-Attar, presumably because Urien et al. used AGP samples from different origins, which led to different binding parameters as will be discussed later in section IV E.

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	Survey of bin	ding parameters for th	TABLE 11 e binding of basic and s	l1 some neutral di	TABLE 11 are to binding parameters for the binding of basic and some neutral drugs with isolated AGP and HSA	ASH b		
Drug	Origin of AGP	FAGE	Free	RAGP	KAOP [M <sup>-1</sup> ]	PHEN	Kuna [m <sup>-1</sup> ]	Ref.
Alfentanil	Miles	AGP concentration dependent, 0.20 → 0.06 (50 → 200 mg/100 ml)	HSA concentration dependent, 0.97 $\rightarrow 0.65 (0.1 \rightarrow 6$ g/100 ml)					348
Alprenolol	Behringwerke, 67 mg/100 ml	0.24	0.77					75
	Behringwerke, 66 mg/100 ml	0.45	0.60		ŝ			42, 43
		$F_{AGP+HBA} = 0.29$ (same range as in	ne range as in m)					
Amitriptyline	Behringwerke		Ì	0.97 and 1.94	Two sites 3.4 × 10 <sup>6</sup> and 1.3 × 10 <sup>4</sup>	6.16	$3.7 \times 10^{4}$	8
	Behringwerke	$\begin{array}{c} 0.165 \\ F_{AOP + HSA} = 0.10 \\ E \\ -0.0 \end{array}$	0.146 \ = 0.10 0.654					82
	Gift from oth-	<i>г</i> адр + нва + цро <b>=</b> 0.004 0.334 0.334	IPo = 0.064 0.365					413
Aprindine	Behringwerke	Drug concentration dependent, 0.13–	0.15-0.18	1.1	Two sites	1	$9.8 \times 10^{3}$	524
		0.73		and 2.9	and 8.3 × 10°			
		$F_{AGP + HBA} = 0.06 - 0.16$ (same range as in serum)	0.06 – 0.16 as in serum)	In solution of 1 0.53	In solution of mixture of HSA and AGP 0.53 1.4 × 10 <sup>6</sup>	66.0	$1 \times 10^{4}$	
Bupivacaine	Own prepara- tion using modifica-			Š	<i>pH dependent</i> <i>pH 7.4 → 7.0</i> 3 (5.3 → 0.146) × 10 <sup>6</sup>	Hd Hg 46	<i>pH</i> independent <i>pH</i> 7.4 → 7.0 .46 (8.98 → 4.87) × 10 <sup>3</sup>	140
	tion of method of Hao and Wickerhau- ser (217)							
	Behringwerke	Concentration de- pendent, 0.14 → 0.69 (60 → 20 mg/100 ml)						\$
	Not men- tioned	0.27						410
	AGP (200 mg/100 ml) added to plasma (!!)		Decrease from 0.	.16 to 0.1 <b>4 after</b>	Decrease from 0.16 to 0.14 after addition of isolated AGP to plasma	to plasma		410
Carbamazepine (neu- tral drug)	Not men- tioned	Concentration dependent, $0.71 \rightarrow 0.90 (150 \rightarrow 50 \text{ mg/100 ml})$	0.32	Agree with tho	Agree with those calculated in plasma			339

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## KREMER, WILTING, AND JANSSEN

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340	551	282	158 363 590	020 264, 255 462	282	Q	442	167	514	281	313	133	315	317
	≧6.6 × 10³			3.0 × 10²						$n_{\rm HBA}K_{\rm HBA} = 5.7 \times 10^4$				
$4 \times 10^{2}$	۲.			6.8						пнвлК				
7.5 → 7.33	Two sites 9.4 × 10 <sup>6</sup>	$5 \times 10^{6}$ $3.4 \times 10^{6}$	$3 \times 10^{6}$ $3.45 \times 10^{6}$	4.7 × 10 <sup>4</sup> 6.3 × 10 <sup>4</sup>	4 × 104 (by drug dis- placement)			Two sites 1.55 × 10 <sup>7</sup> and <b>4</b> × 10 <sup>6</sup>	$6.25 \times 10^{\circ}$	$8 \times 10^{6}$	1.0 × 10 <sup>6</sup>	Two sites 8.84 × 10 <sup>6</sup> and 2.43 × 10 <sup>4</sup>	Stereoeelective $R(-) = 5.12 \times 10^{\circ}$ $S(+) = 8.9 \times 10^{\circ}$ Racemic = $6.2 \times 10^{\circ}$	$9.5 \times 10^{\circ}$
1.7 × 10°	0.5 and	1	0.83	1.3 1				0.9 and 0.9	1	1	0.2	0.256 and 0.606		0.02(!!)
1.44 → 1.36, 0.65 → 0.52 (4 → 2 g/ 100 ml)	0.20			0.38			HSA concentration dependent, 0.013 → 0.024 (5 → 2.3 g/100 ml)					Drug concentration dependent, 0.81– 0.98		
Concentration dependent, $0.66 \rightarrow 0.88 (100 \rightarrow 50 \text{ mg/100 ml})$	Drug concentration dependent, 0.08- 0.60	3	0.076 0.95	0.32		0.81	Decreases after ad- dition of AGP to HSA solution from 0.0156 → 0.0109				Drug concentration dependent, 0.14– 0.75	Drug concentration dependent, 0.35- 0.97 AGP concentration dependent, 0.15 → 0.75 (140 → 40 mg/100 ml)	i	
Concentration dependent	Sigma	Miles, defat- tad	Behringwerke	Milles Calbiochem Behringwerke	Own prepara- tion and de- fatted by	Not men-	Not men- tioned	Behringwerke	Own prepara- tion	Own prepara- tion	Sigma	Not men- tioned	Sigma	Sigma (600 mg/100 ml!!)
Sigma	Chlorpromazine		lok-i-ols:V	Ciclazzingoi Desmethylimipramine Diazepam (neutral drust!)				Dipyridamole			Disopyramide			
	ПЛ			XX				x			ХІІ			

DRUG BINDING TO HUMAN AGP IN HEALTH AND DISEASE

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By AGP-constrained         Dependent on HSA         (9 × 10 <sup>3</sup> ) - (2 × 10 <sup>3</sup> )           Restanded         ESA samples, action ASR to a Station         Dependent on HSA         (9 × 10 <sup>3</sup> ) - (2 × 10 <sup>3</sup> )           Addition         Decrease of free fraction after AOP addition ACP to a solution action a	Drug	Origin of AGP	FAGP	FHRA	NAGP	К <sub>АФР</sub> [M <sup>-1</sup> ]	лнал	KHBA [M <sup>-1</sup> ]	Ref.
HSA and the fraction after AGP addition HSA and the fraction after AGP addition Addition Person of the fraction after AGP addition Addition (CP constration atter AGP addition (CP constration atter AGP addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to pharma decreases free fraction from 0.129 to 0.064 Addition (CP to 0.01 to		By AGP-con-		Dependent on		Dependent on HSA	(3)	3 × 10²) – (2 ×	
pier pier turna Affino of Demensor free fraction after ACIP addition turna Affino of Point Construction affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.139 to 0.064 affinition of ACP to planna decreases free fraction from 0.130 to 0.064 affinition of ACP to 0.01 affinition of ACP to planna decreases free fraction from 0.130 to 0.04 affinition of ACP to planna decr		taminated HSA sam-		HSA samples, 0.95-0.67		samples (1.7–3.3) × 10 <sup>6</sup>		10°)	
Addition of Decrease of free fraction after AGP addition during the addition of the fraction after AGP addition addition of Ad		ples				ł			
Minum Minum AGP and AG concentration a value a value		Addition of	Decrease of free frac	tion after AGP additic	n				244
<ul> <li>monto activity and activity of a consentration activity of a consentration activity of a consentration activity activity</li></ul>		human ACD to							
Behrugeneta (appendent, 0.13)         10           Behrugeneta (appendent, 0.13)         0.61           Outhout         0.01           Behrugeneta (appendent, 0.53)         0.91           Ormanor- vereta)         0.17         Addition of AGP to plaama decreases free fraction from 0.126 to 0.064           Ann prepare tion and de- tion		rabbit							
Bultingwette         APF concentration         10 $= 0.67$ (2001) $= 0.67$ (2001) $= 0.67$ (2001) $= 0.67$ (2001) $= 0.67$ (2001) $= 0.67$ (2001) $= 0.67$ (2001) $= 0.68$ (1001) $BA$ (100 mi) $= 0.081$ (2001) $= 0.023$ (50 - 3 g/ $= 0.023$ (50 - 3 g/ $= 0.081$ (2001) $= 0.023$ (50 - 3 g/ $= 0.023$ (50 - 3 g/ $= 0.081$ (100 mi) $0.01$ $0.01$ $1$ $0.01$ $0.01$ $0.01$ $1$ $0.01$ $0.01$ $0.01$ $1$ $3.5 \times 10^{4}$ $0.01$ $0.01$ $1$ $3.5 \times 10^{4}$ $0.064$ $0.01$ $0.01$ $1$ $3.5 \times 10^{4}$ $3.6 \times 10^{4}$ $0.01$ $0.01$ $0.01$ $1$ $3.5 \times 10^{4}$ $3.5 \times 10^{4}$ $0.01$ $0.01$ $0.01$ $1$ $3.5 \times 10^{4}$ $3.5 \times 10^{4}$ $0.01$ $0.01$ $0.01$ $1$ $3.5 \times 10^{4}$ $3.5 \times 10^{4}$ $0.01$ $0.01$ $0.01$ $0.01$ $0.01$ <		serum							
Signal         Computer of consentration decretation         RA concentration decretation         RA concentration dependent, 0.46         RA concentration dependent, 0.46           nag/100 ml)         0.07         0.023 (6 - 3 g/ 0 nag/100 ml)         0.033         1         3.5 × 10 <sup>4</sup> nag/Kiaa, = 1 × 10 <sup>5</sup> Remain         0.07         0.07         0.031         1         3.5 × 10 <sup>4</sup> nag/Kiaa, = 1 × 10 <sup>5</sup> Remain         0.07         0.017         Addition of ACP to plasma decreases free fraction from 0.128 to 0.064         naa, Kiaa, = 1 × 10 <sup>5</sup> Not men         0.17         Addition of ACP to plasma decreases free fraction from 0.128 to 0.064         naa, Kiaa, = 1 × 10 <sup>5</sup> Not men         0.17         Addition of ACP to plasma decreases free fraction from 0.128 to 0.064         1           Not men         0.17         Addition of ACP to plasma decreases free fraction from 0.128 to 0.064         1           Not mend de- tion and de- tion and de- fated by         0.02 (6 - 0.11)         2.6 × 10 <sup>6</sup> (6) (9) (9)         1           Orn prepare         0.03         0.01 (9) drug dis- placement(1)         2.6 × 10 <sup>6</sup> (9) (9) drug dis- tion and de- fated by         2.8 × 10 <sup>6</sup> (9) drug dis- placement(1)         2.8 × 10 <sup>6</sup> (9) drug dis- placement(1)         2.8 × 10 <sup>6</sup> (9) (9) drug dis- tion and de- fated by         1         9.2 × 10 <sup>6</sup> (9)         1 <tr< td=""><td></td><td>Behringwerke</td><td>AGP concentration dependent, 0.19 <math>\rightarrow 0.67 (200 \rightarrow 0.57 (200 \rightarrow 0.57))</math></td><td>1.0</td><td></td><td></td><td></td><td></td><td>78</td></tr<>		Behringwerke	AGP concentration dependent, 0.19 $\rightarrow 0.67 (200 \rightarrow 0.57 (200 \rightarrow 0.57))$	1.0					78
in Behringwerke 0.38 dependent, 0.46 $-0.061 (10-3) = 1.066 (10-3) = 0.061 (10-3) = 0.061 (10-3) = 0.01 0 m)$ wealth $0.01 = 0.01 = 0.01 = 0.01$ $0.01 = 0.001$ $0.01$ $0.00000000000000000000000000000000000$	kepin	Sigma	AGP concentration	HSA concentration					552
in Behringwerke $0.45$ $0.31$ 1 $3.5 \times 10^4$ $n_{max} K_{max} = 1 \times 10^6$ Common- $0.07$ $0.07$ $0.11$ Addition of ACP to plasma decreases free fraction from 0.128 to 0.064 Not men- $0.17$ Addition of ACP to plasma decreases free fraction from 0.128 to 0.064 tioned ACP concentration HSA concentration dependent, $0.36$ $-0.01$ $2.6 \times 10^6$ (by drug dis- $0.0^{m}$ prepares $0.07$ $-0.06$ ( $150 \rightarrow -0.08$ ( $100  ml$ )) $0.0^{m}$ prepares $0.01$ $2.6 \times 10^6$ (by drug dis- tota and de- trated by charcool Behringwerke $0.31$ $0.54$ $0.11$ $2.6 \times 10^6$ (by drug dis- tota and de- tota and de- tota and de- tota and de- tota and de- tota and de- Behringwerke $0.31$ $0.54$ $0.11$ $2.6 \times 10^6$ $0.2$ $2.3 \times 10^6$ $0.5$ $0.46$ $0.11$ $9.2 \times 10^6$ $0.2$ $1.1 \times 10^6$ tota and de- tota an			dependent, 0.28 → 0.69 (120 → 30 mg/100 ml)	dependent, 0.46 → 0.62 (5 → 3 g/ 100 ml)					
Common- Serum Lab- oustory Serum Lab- oustory Normer 103 Serum Lab- oustory Normer 1035 dependent, 0.55 -0.07 (150	ythromycin	Behringwerke	0.45	0.91	1	$3.5 \times 10^{4}$	$n_{ m HSA}K_{ m HS}$	$h = 1 \times 10^2$	420
weith Serun Lab. or men- or tory Not men- or tory Not men- tioned Not men- tioned Miles AGP concentration Miles AGP concentration Miles AGP concentration AGP concentration Miles AGP concentration AGP concentration dependent, 0.35 $\rightarrow 0.70$ (130 $\rightarrow -0.36$ (6 $\rightarrow 0.1$ apendent, 0.35 $\rightarrow 0.70$ (130 $\rightarrow -0.36$ (6 $\rightarrow 0.1$ apendent, 0.50 $\rightarrow 0.70$ (130 $\rightarrow -0.36$ (6 $\rightarrow 0.1$ apendent, 0.50 $\rightarrow 0.70$ (130 $\rightarrow -0.36$ (6 $\rightarrow 0.1$ apendent, 0.50 $\rightarrow 0.70$ (130 $\rightarrow -0.36$ (6 $\rightarrow 0.1$ apendent, 0.50 apendent, 0.50 apendent, 0.50 apendent, 0.50 apendent, 0.50 approximation $approximation approximationapproximation approximation approximat$	idocaine	Common-	0.07						357
Oritory to men- to trans- to trans- to trans- to trans- to trans- mining     O 17     Addition of AGP to plasma decreases free fraction from 0.128 to 0.064       Miles     AGP concentration dependint, 0.35     Addition of AGP to plasma decreases free fraction from 0.128 to 0.064       Miles     AGP concentration dependint, 0.35     Approximation dependint, 0.56       Own prepara- tion and de- tion and de- tion and de- tion recoal     2.6 × 10° (by drug dia- placement)       Behningwerke     7.0 sites       Own prepara- tion and de- tiated by charcoal     2.8 × 10° (by drug dia- placement)       Own prepara- tion and de- fatted by charcoal     2.8 × 10° (by drug dia- placement)       Own prepara- tion and de- fatted by charcoal     0.11     9.2 × 10° and befatted 2.8 × 10° befatting 1.5 ×       Own prepara- tion and de- fatted by fatted by fatted by     0.54     1.1     9.2 × 10°       Own prepara- tion and de- fatted by     0.46     1.1     9.2 × 10°       Own prepara- tion and de- fatted by     0.15     0.46     1.1		wealth Serum Lab-							
Niles ACP concentration tioned Miles ACP concentration HSA concentration dependent, 0.35 dependent, 0.50 $\rightarrow 0.70 (150 \rightarrow -0.50 (150 \rightarrow -0.15 0.0004)$ fatted by fatted by		oratory		- 17:66 4			, 0 100 J		
Mile     AGP concentration     HSA concentration       dependent, 0.35     dependent, 0.36     dependent, 0.50       → 0.70 (150→     → 0.36 (6 → 0.1       0 mg/100 ml)     g/100 ml)       50 mg/100 ml)     g/100 ml)       50 mg/100 ml)     g/100 ml)       6 mont de- tated by charcoal     - 0.36 (6 → 0.1       0 mg/100 ml)     g/100 ml)       10 0 ml     g/100 ml)       2.6 × 10 <sup>4</sup> dug dis- placement)       10 0 ml prepara- tion and de- fatted by charcoal     - 0.36 (6 → 0.1       0 ml prepara- tion and de- fatted by     2 and positive     7.5 × 10 <sup>4</sup> dug dis- placement)       0 ml prepara- tion and de- fatted by     0.46     1.1     9.2 × 10 <sup>4</sup> 0 mont esters     0.46     1.1     9.2 × 10 <sup>4</sup> 0 mont esters     1.3     0.46     1.1 × 10 <sup>5</sup> 0 mont esters     0.46     1.1     9.2 × 10 <sup>4</sup> 0 mont esters     1.1     9.2 × 10 <sup>4</sup> 0 mont esters     1.1 × 10 <sup>5</sup>		not men- tioned	11.0	AGGIUG	n oi AGP to plat	sima decreases iree iraction	01 02 07.17.0 10.17.0 10	1.064	410
the over the content, 0.35 dependent, 0.50 $\rightarrow 0.36 (6 \rightarrow 0.1)$ 50 mg/100 ml) g/100 ml)	ntanil	Miles	AGP concentration	HSA concentration					348
e     Own prepare tion and de- fatted by charcoal     2.6 × 10 <sup>6</sup> (by drug dis- placement)       Behringwerke     2.8 × 10 <sup>6</sup> (by drug dis- placement)       Own prepare- tion and de- fatted by charcoal     2.6 × 10 <sup>6</sup> (by drug dis- placement)       Own prepare- tion and de- fatted by charcoal     2.8 × 10 <sup>6</sup> (by drug dis- placement)       Definitigwerke     0.31       0.12     0.46       0.12     0.46       0.12     0.46       1.1     9.2 × 10 <sup>6</sup> be- britted by       fatted by charcoal     0.5       Defatted 2.8 × 10 <sup>6</sup> be- fatted by     0.2       0.12     0.5       0.12     0.5       0.11     9.2 × 10 <sup>6</sup> be- britted by       0.11 be- tion and de- fatted by     0.5			dependent, 0.35 → 0.70 (150 → 50 ms/100 ml)	dependent, 0.50 $\rightarrow 0.95 \ (6 \rightarrow 0.1$ $g/100 \ ml)$					
tion and de- fatted by charcoal Behringwerke fatted by concentivity cooperativity own prepara- tion and de- fatted by charcoal Behringwerke 0.31 0.54 Calbiochem 0.12 0.46 fatted by fatted by fatte	uphenazine	Own prepara-				$2.6 \times 10^{6}$ (by drug dis-			282
charcoal Behringwerke Tuo sites Behringwerke 7.5 × 10° and positive cooperativity cooperativity Own preparation and de- tion and de- fatted by charcoal 0.31 0.54 Behringwerke 0.31 0.54 Calbiochem 0.12 0.46 1.1 9.2 × 10° be- 0.2 1.1 × 10° fatted by charcoal 0.12 0.46 1.1 9.2 × 10° be- 0.2 1.1 × 10° bigher drug 1.5 × 10° be- 0.2 1.1 × 10° bigher drug 1.5 × 10° be- 0.2 1.1 × 10° bigher drug 1.5 × 10° be- 0.2 1.1 × 10° bigher drug 1.5 × 10° be- 0.2 1.1 × 10° bigher drug 1.5 × 10° be- 0.2 1.1 × 10° brated by bigher drug 1.5 × 10° bigher drug 1.5 × 10° be- 0.2 1.1 ×		tion and de- fatted hv				placement)			
Behringwerke       Two sites         0wn preparation       2 and positive         100 and deficient       2 and positive         101 and deficient       0.31         1012       0.54         11       9.2 × 10°         11       9.2 × 10°         11       9.2 × 10°         11       9.2 × 10°         11       9.2 × 10°         11       9.2 × 10°         11       9.2 × 10°         11       9.2 × 10°         11 bit       9.2 × 10°         11 calbiochem       0.1         11       9.2 × 10°         11 calbiochem       0.2		charcoal							
Own preparativity     cooperativity     cooperativity       tion and deter     fatted by       fatted by     6 × 10° (by drug displacement)       fatted by     placement)       charcoal     0.31     0.54       Behringwerke     0.31     0.54       Calbiochem     0.12     0.46     1.1       Own preparativity     0.12     0.46     1.1       Own preparative     0.2     1.1     9.2 × 10°       fatted by     fore defatting 1.5 ×     1.1 × 10°       fatted by     higher drug     10°       charcoal     0.0     1.1	Haloperidol	Behringwerke				Two sites $7.5 \times 10^3$ and positive			462
Own preparationOwn preparationtion and detection6 × 10° (by drug dustriated by fatted by charcoalfatted by charcoal0.310.120.54Behringwerke0.310.120.461.19.2 × 10°0.120.461.19.2 × 10°0.120.461.19.2 × 10°0.120.461.19.2 × 10°0.120.460.120.55Defatted 2.8 × 10° be-0.21.1 × 10°fatted by1.1 × 10°fatted by1.1 × 10°charcoal0.00.001.10.001.1		¢			cooperativity	cooperativity			
fatted by fatted by charcoal Behringwerke $0.31$ $0.54$ Calbiochem $0.12$ $0.46$ $1.1$ $9.2 \times 10^6$ $7$ $2.3 \times 10^2$ Calbiochem $0.12$ $0.46$ $1.1$ $9.2 \times 10^6$ be- tion and de- tion and de- fatted by fatted by fatted by for there is a fore defatting $1.5 \times$ here is a fore defatting $1.5 \times$ here is a fore defatting $1.5 \times$ by the is a fo		Uwn prepara- tion and de-				6 × 10° (by drug dis- nlacement)			482
charcoal Behringwerke 0.31 0.54 Behringwerke 0.31 0.54 Calbiochem 0.12 0.46 1.1 9.2 $\times 10^{4}$ 7 2.3 $\times 10^{2}$ Own prepara- 0.12 0.46 1.1 $\times 10^{6}$ Own prepara- 0.12 0.46 1.1 $\times 10^{6}$ finn and de- $0.2$ 1.1 $\times 10^{6}$ fistted by fatted by higher drug 10 <sup>6</sup> higher drug 10 <sup>6</sup> hevels 0.000 1.1 $\times 10^{6}$		fatted by							
Demingwerke $0.01$ $0.04$ Calbiochem $0.12$ $0.46$ $1.1$ $9.2 \times 10^4$ $7$ $2.3 \times 10^2$ Calbiochem $0.12$ $0.46$ $1.1$ $9.2 \times 10^4$ $7$ $2.3 \times 10^2$ Carbiochem $0.12$ $0.46$ $1.1$ $9.2 \times 10^4$ $7$ $2.3 \times 10^2$ Own preparation and determine $0.12$ $0.46$ $1.1$ $9.2 \times 10^4$ $7$ $2.3 \times 10^2$ Carbon preparation and determine $0.12$ $0.46$ $1.1$ $0.5$ Defatted $2.8 \times 10^4$ between the state of the s		charcoal D-L-:	10.0	0.64					ł
0.12 0.46 1.1 9.2 × 10 <sup>4</sup> 7 2.3 × 10 <sup>2</sup> 0.5 Defatted 2.8 × 10 <sup>6</sup> be- 0.2 1.1 × 10 <sup>6</sup> More sites at fore defatting 1.5 × higher drug 10 <sup>6</sup> levels	upramine	Calbiochem	0.19	0.0 <del>4</del> 0.46					755
Own preparation and determined     0.5     Defatted 2.8 × 10 <sup>6</sup> be-     0.2     1.1 × 10 <sup>6</sup> tion and determined     More sites at fore defatting 1.5 ×     fatted by     ingher drug     10 <sup>6</sup> charcoal     10 <sup>6</sup> levels     10 <sup>6</sup> levels     10 <sup>6</sup>		Calbiochem	0.12	0.46	1.1	$9.2 \times 10^{4}$	7	$2.3 \times 10^2$	254
tion and de- fatted by charcoal levels 1.5 × b. f. charcoal 10° b. f. charcoal 10°	iipramine	Own prepara-			0.5	Defatted 2.8 $\times$ 10 <sup>6</sup> be-	0.2	$1.1 \times 10^{6}$	282
higher drug 10 <sup>5</sup> levels 1000000000000000000000000000000000000		tion and de-			More sites at	fore defatting 1.5 ×			
		fatted by			higher drug	10 <sup>6</sup>			
		CDAFCOAL	000		si i				000

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26



462		135	197	410	411			282	7			201	400	9	42, 43	367	8	242	2	491			433								
				n from 0.29 to 0.15 nn from 0.36 to 0.14	Addition of AGP to cord serum decreases free fraction from 0.44 to 0.25 which is in same range as found in serum of	n 0.3 to 0.8			$n_{\rm HSA}K_{\rm HSA} = 3.2 \times 10^2$					Addition of isolated AGP to plasma or a solution of HSA decreased free fraction with increasing AGP concentration						Two sites		2.3	1.3 $6.1 \times 10^3$								
Two sites	2.2 × 10° and 4 R × 10°			Addition of AGP (50 mg/100 ml) to serum decreases free fraction from 0.29 to 0.15 Addition of AGP (200 mg/100 ml) to plasma decreases free fraction from 0.36 to 0.14	from 0.44 to 0.25 which is ir	adults Removal of AGP from plasma increases free fraction from 0.3 to 0.8	•	2.4 × 10° (by drug dis- placement)	Two sites	$4 \times 10^{\circ}$	and 69 × 10 <sup>2</sup>			A decreased free fraction wi						Two sites	pue	$3.37 \times 10^{3}$	$1.84 \times 10^{4}$		$2.16 \times 10^{\circ}$	$1.12 \times 10^{4}$		101 0 100	01 × 100		
	1 and 9	9		00 ml) to s 00 ml) to p	e fraction f	om plasma				0.38	and 8.4	5		tion of HS <sup>1</sup>						1 05	uz.1	5.8	0.94		0.94	0.94			0.94		
		HSA concentration dependent, 0.72 → 0.79 (5 → 3 g/ 100 ml)	0.77	ldition of AGP (50 mg/1 lition of AGP (200 mg/1	ord serum decreases fre	Removal of AGP fr			0.84				0.64	AGP to plasma or a solu	0.85	wer than in serum	0.82	0.89		Drug concentration	dependent, v.vo										
		AGP concentration dependent, 0.79 → 0.90 (100 → 50 mz/100 ml)		Add	Addition of AGP to c				0.2				AGF concentration dependent, $0.09$ $\rightarrow 0.92 (200 \rightarrow$ 67  mg/100 ml)	Addition of isolated /	0.98	$F_{AGP + HSA} = 0.80$ , lower than in serum	AGF concentration dependent, 0.80 $\rightarrow 0.96 (60 \rightarrow 20$ $m_{\sigma}(100 ml)$	0.68		Drug concentration	dependent, U.20	$\downarrow 6 \times 10^{-4} \text{ M}$									
Behringwerke	)	Sigma	Gift from	others AGP added to	plasma AGP added to	cord serum Plasma defi- cient in	AGP (!!)	Own prepara- tion	Sigma			(	Own prepara- tion		Behringwerke		Behringwerke	Gift from oth-	ers	Sigma			Own prepara-	tion	Desialylated	Carhorv-	methylated	AGP	Desialylated	and carboxy- methylated AGP	->
		Ketamine	Lidocaine					Loxapine	Methadone						Metaprolol		Meperidine			Mianserin			Nicerzoline								
		XX	ХХІ					ΪХХ	IIIXX						XXIV		XXX			ΙΛΧΧ			ΙΙΛΧΧ								

## DRUG BINDING TO HUMAN AGP IN HEALTH AND DISEASE

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28

				TABLE 11-Continued	Continued				
No.	Drug	Origin of AGP	From	FIREA	RAGE	K <sub>AGP</sub> [M <sup>-1</sup> ]	A America []	K [m <sup>-1</sup> ]	Ref.
IIIAXX	Nortriptyline	Behringwerke				Two sites	5.2 4.4	$4.4 \times 10^{2}$	8
					0.97	$1.2 \times 10^{6}$			}
					0.87	2 × 10			
		Behringwerke	0.32	0.42					82
			FAGP + HSA =	0.185					
XIXX	Ornenolol	Rehringwerke	0.28 0.78	0.78					42.43
			5	- 0.19					r F
			$r_{AOP} + HBA = 0.12$	21.0					
		Sigma	0.25	0.80		Two sites			\$
					0.2 and not	$1.9 \times 10^6$ and not			
					neuomen	neuonueu	:	•	
XXX	PCR 2362	Behringwerke		0.18	es es	$3.3 \times 10^{\circ}$	$n_{\rm HBA}K_{\rm HBA} = 7.1 \times 10^3$	× 10'	196
XXXI	Perazine	Behringwerke				Two sites	Two sites		8
					0.07	401 × 6 L	c 7	1 0 \ 108	}
					10.0	$01 \sim 7.1$		۰ IC	
					and	and	and	and	
					3.9	$7.4 \times 10^{3}$		$4.1 \times 10^{2}$	
		Not men-				$3.8 \times 10^6$ (by drug dis-			460
		tioned				placement)			
		Bahrinmerka	0.060	111					ŝ
		Detrutingworke		111.0					70
			$F_{AGP} + HSA = 0.040$	c#0.04					
			Same as in serum	serum					
		Behringwerke				Two sites			462
						$3.8 \times 10^6$			
					and	and			
					4	4 2 X 10 <sup>3</sup>			
			<b>5</b> 50	92.0	•				001
IIYYY	Pnencycuame		0.00	0.0					0A1
		919 							
		Sigma	AGP concentration	0.81		$1.74 \times 10^{\circ}$			390
			dependent, 0.90						
			↓ 0.60 (200 ↓						
					•				
		i	$F_{AGP+HBA} = 0.25 \rightarrow 0.54$	20 <b>+</b> 0.04	In mixtur	In mixture of AGP and HSA enhanced affinity to AGP, $7.72 \times 10^{\circ}$	d attinity to AGP, 7.72	$2 \times 10^{\circ}$	
		Sigma	0.87	0.87					31
		Calbiochem							
IIIXXX	Phenothiazin deriva-	Not men-				For different deriva-			<b>4</b> 60
	tives	tioned				tives between (0.4-			
						$11) \times 10^4$ (by drug			
						displacement)			
XXXIV	Perohenazine	Sigma	Drug concentration	0.27		two sites	$n_{\rm HSA}K_{\rm HSA} = 4.6 \times 10^3$	× 10³	551
		8	denendent	•	0.5	$3.4 \times 10^{6}$			4000
			0.15 (//10 <sup>8</sup> mc/ml)		ere d				
			$\frac{1}{2}$		<b>1.</b> 0				•
XXXV	Pindolol	Behringwerke	0.70	0.82					42, 43
			$F_{AGP+HBA} = 0.44$ , same as in serum	me as in serum					
		Behringwerke	Drug concentration	0.82	1.4	$7.1 \times 10^{4}$			305
			dependent						
			$0.35 (< 5 \times 10^3$						
			ng/ml)						
			0.60 (>5 × 103						

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	305	200	181	182	567	582	551	41, 42	194	282	11	2	363 411	582
•									лналКнал = 1.8 × 10 <sup>3</sup>				in serum of adults	
	In the range of (10 <sup>4</sup> – 10 <sup>6</sup> )		Type $c$ 9.4 × 10°	Type $p = type c$ delipidated 0.9 10.8 × 10 <sup>6</sup> Type $p$ recombined with lipid from type $c$ 0.17 8.5 × 10 <sup>6</sup> Ethanolic precipitation 1 8 × 10 <sup>6</sup>		$3.2  imes 10^{6}$			n <sub>aap</sub> K <sub>aap</sub> = 3.05 × 10 <sup>4</sup>	1 × 10° (by drug dia- placement)		Two sites 2.44 × 10 <sup>6</sup> and not mentioned	$1.05$ $1.13 \times 10^{6}$ Increase of binding in fetal serum after addition of AGP to same range as in serum of adults	Native 8.4 × 10 <sup>5</sup> Desialylated 6 × 10 <sup>5</sup>
			0.2	Type I 0.9 7ype p recom 0.17 Eth		1			<b>N</b> AGF			0.6 and not mentioned	1.05 serum after add	1 1
		60.0			0.18		0.53	0.45 same as in serum	Drug concentration dependent, 0.45– 0.53		0.50	0.60	e of binding in fetal s	
		0.35			0.10	Desialylation, no effect; 0.15	Drug concentration dependent 0.17(<10 <sup>2</sup> mg/ml) 0.4 (>10 <sup>3</sup> mg/ml)	0.31 0.45 0.45 0.41. same as in serum	Drug concentration dependent, 0.40– 0.95		Stereceelective $F^{-}_{AOP} = 0.23^{\circ}$ $F^{+}_{AOP} = 0.30$ $F^{-}_{AOP} = 0.26$	0.30	0.21 Incream	Drug concentration dependent, 0.05 → 0.17 (0.4 → 2 × 10 <sup>-4</sup> M)
	Behringwerke	Not men- tioned	Own prepara- tion using chromatog- raphy (type c) or ethan- olic meth- ods		Gift from oth- ers	Sigma, native and desialy- lated	Sigma	Behringwerke	Behringwerke	Own prepara- tion and de- fatted by charcoal	Sigma	Sigma	Behringwerke Addition of AGP to fe- tal serum	Sigma, native and desialy- lated
)	Pindolol-related com-	Prazosin	Progesterone (neutral drug!!)				Propranolol				Propranolol			
	ΙΛΧΧΧ	ΠΛΧΧΧ	IIIAXXX				XIXXX							

## DRUG BINDING TO HUMAN AGP IN HEALTH AND DISEASE

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29

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30

	Ref.	415	172	282	. 381	282	•	067	<b>561</b>	4
	Кнал [m <sup>-1</sup> ]				$4.9 \times 10^{4}$		<sup>пнв</sup> Kнва = 9.4 × 10 <sup>3</sup>		лналКнал = 3.3 × 10 <sup>4</sup>	
	NHBA						RHB		Лна	
	К <sub>АОР</sub> [м <sup>-1</sup> ]			8 × 10° (by drug dis- placement)	$6.3 \times 10^{7}$	2.4 × 10° (by drug dis- placement)	8.9 × 10'	Increase of drug binding in serum with increasing AGP added	Two sites $6 \times 10^{\circ}$ and $2 \times 10^{\circ}$	2 × 10° and two weaker sites
tinued	RADP				0.93		က	ıg binding in	1 and 3	
TABLE 11—Continued	FHSA	0.61	me range as serum ion in plasma after ing on amount	8			0.12 0.93 ame as in serum	Increase of dru	0.08	
	FAGP	0.47	$F_{AGP}$ + HEA = 0.23, same range as serum Decrease of free fraction in plasma after AGP addition depending on amount	0000			0.12 0.73 0.93 $F_{AGP + HBA} = 0.44$ , same as in serum		Drug concentration dependent 0.03 (<10 <sup>3</sup> ng/ml) 0.20 (>10 <sup>3</sup> ng/ml)	
	Origin of AGP	Not men- tioned	AGP added to plasma	Own prepara- tion and de- fatted by	charcoal Hoechst Swedish	Own prepara- tion and de- fatted by	Behringwerke Behringwerke	AGP added to serum	Sigma	Sigma
	Drug	Quinidine		Thioridazine		Thiothixene	Ticlopidine Timolol	Triazolam	Trifluoperazine	KLVII DAPN (fluorescent Sigma probe!!)
	No.	XL		ХЦ		IIIX	XILV	ATX	IVLX	XLVII

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Drag	F*	F		K [M <sup>-1</sup> ]	Ref.
Diug	T'AOP	T. Harv	/•A	AP ANAD (m )	1001.
Folic acid	0.97	0.77			249
Indomethacin	0.40	0.09	0.3	$1.86 \times 10^{6}$	
Methotrexate	0.99	0.60			249
Phenylbutazone	0.73	0.07	1.7	$5.25 \times 10^{3}$	249
Phenytoin	0.67	0.34			249
Probenecid	0.99	0.08			249
Retinoic acid	0.67	0.58			249
Sulfinpyrazone	0.70	0.17	0.6	$2.40 \times 10^{3}$	249
Tolmetin	0.99	0.33			249
Acenocoumarol	0.15		1.08	$2.01 \times 10^{5}$	545
Benoxaprofen	0.90				545
Indomethacin	0.90				545
Itanoxone	0.90				545
Phenylbutazone	0.74		0.71	$3.50 \times 10^{4}$	545
Warfarin	0.12		1.09	$2.12 \times 10^{5}$	545
Phenobarbital				Two sites	462
	Indomethacin Methotrexate Phenylbutazone Phenytoin Probenecid Retinoic acid Sulfinpyrazone Tolmetin Acenocoumarol Benoxaprofen Indomethacin Itanoxone Phenylbutazone Warfarin	Folic acid0.97Indomethacin0.40Methotrexate0.99Phenylbutazone0.73Phenytoin0.67Probenecid0.99Retinoic acid0.67Sulfinpyrazone0.70Tolmetin0.99Acenocoumarol0.15Benoxaprofen0.90Indomethacin0.90Itanoxone0.90Phenylbutazone0.74Warfarin0.12	Folic acid         0.97         0.77           Indomethacin         0.40         0.09           Methotrexate         0.99         0.60           Phenylbutazone         0.73         0.07           Phenylbutazone         0.73         0.07           Phenylbutazone         0.67         0.34           Probenecid         0.99         0.08           Retinoic acid         0.67         0.58           Sulfinpyrazone         0.70         0.17           Tolmetin         0.99         0.33           Acenocoumarol         0.15           Benoxaprofen         0.90           Indomethacin         0.90           Phenylbutazone         0.74           Warfarin         0.12	Folic acid         0.97         0.77           Indomethacin         0.40         0.09         0.3           Methotrexate         0.99         0.60           Phenylbutazone         0.73         0.07         1.7           Phenylbutazone         0.67         0.34         9           Probenecid         0.99         0.08         8           Retinoic acid         0.67         0.58         5           Sulfinpyrazone         0.70         0.17         0.6           Tolmetin         0.99         0.33         4           Acenocoumarol         0.15         1.08         5           Benoxaprofen         0.90         1         1           Indomethacin         0.90         1         1           Phenylbutazone         0.74         0.71         0.71	Folic acid         0.97         0.77           Indomethacin         0.40         0.09         0.3         1.86 × 10 <sup>6</sup> Methotrexate         0.99         0.60             Phenylbutazone         0.73         0.07         1.7         5.25 × 10 <sup>3</sup> Phenylbutazone         0.73         0.07         1.7         5.25 × 10 <sup>3</sup> Probenecid         0.99         0.08             Retinoic acid         0.67         0.58             Sulfinpyrazone         0.70         0.17         0.6         2.40 × 10 <sup>3</sup> Tolmetin         0.99         0.33             Acenocoumarol         0.15         1.08         2.01 × 10 <sup>6</sup> Benoxaprofen         0.90              Indomethacin         0.90              Phenylbutazone         0.74         0.71         3.50 × 10 <sup>4</sup> Warfarin         0.12         1.09         2.12 × 10 <sup>5</sup>

TABLE 12

Survey of binding parameters reported for the binding of acidic drugs

with AGP

\*  $F_{AGP}$ , free fraction in AGP solution (200 mg/100 ml in ref. 249 and 90 mg/100 ml in ref. 545, respectively);  $F_{HBA}$ , free fraction in HSA solution (4.5 g/100 ml).

1 and not 8

mentioned

× 10<sup>2</sup>

cooper-

ativity

and positive

Not included in table 12 are recent results on the binding to AGP of the acidic drug apazone (544).

## D. Nature and Number of Binding Sites on Alpha-1acid Glycoprotein

Several models are used to calculate the number of binding sites and the binding constant (table 13, no. VIII; section IV, A and E). From the literature it follows that in some cases a particular method can result in an incorrect estimate of the binding parameters (1, 40, 417). Therefore, care should be taken when using a particular analyzing procedure. It is advisable to use a computerized method (365).

The differences between the binding-parameters obtained in plasma and obtained in a solution of isolated pure plasma proteins (tables 8, 10–11), as reported by several authors for a defined drug, can be ascribed to the different methods of interpretation and the different origins of the plasma protein fractions used. It should be noted that, in studies dealing with the binding in plasma, authors generally use a model with only one class of binding site(s) for each plasma protein, whereas in studies dealing with the binding in solutions of an isolated plasma protein, they often use a model with two classes of binding sites on one plasma protein.

From the result of the drug displacement studies done with isolated AGP, it can be concluded that on AGP there is only one common binding site for all the basic drugs studied (4, 158, 197, 363, 364, 463) and that the amounts of drug displaced can be correlated with the binding constant. Although this indirect method points to only one single binding site on AGP, other direct methods, such as the curvilinear Scatchard plots, indicate the presence of more than one class of binding sites on AGP (table 11).

The nature of the drug binding to AGP has been the subject of several studies (157, 158, 160, 177, 181, 182, 305, 348, 364, 460-465, 555, 582). The number of reported binding sites on AGP (summarized partly in table 11) can increase to 7, as will be discussed further below. For progesterone and other steroid hormones (181) the number of binding sites on AGP depended on the isolation method used (table 11). Chromatographically purified AGP had a lower binding capacity than alcohol-precipitated AGP or delipidated AGP. Interaction is weakest for the steroids with the highest polarity. The binding of progesterone seems to alter with the conformation of AGP, depending on the neutral salt used (182). Ganguly and Westphal (182) reported on a conformational change of AGP from a more compact globular protein structure towards a random coil, resulting in the reduced availability of the hydrophobic residues near the binding site on AGP. The results of Wallace and Halsall (555) support the hydrophobic nature of the binding of progesterone to AGP. Kerkay and Westphal (272) found that the  $\Delta^4$ -3-keto group (566) is involved in the interaction of steroids with AGP; they also found that progesterone, cortexone, cortisol, and testosterone associated with AGP at a single primary binding site, whereas estradiol interacted even at 7 and 3 binding sites, respectively, depending on the temperature. Using photoaffinity labeling, Wallace and Halsall (555) found (a) that the residues Lys 162 and Glu 136, 140, or 143 were modified and that these modified residues lie at the periphery of a domain whose central region is hydrophobic and (b)that the existence of Cys-Cys at 72-165 and the modification of Lys 162 are in agreement with the extreme sensitivity of the progesterone binding to disulfide bridge perturbants. Shami et al. (491) found two classes of binding sites for mianserin on AGP and in total seven binding sites, two of which had a high affinity. Perazine was shown to have one site with a high affinity and about four with a lower affinity (83, 462). Trifluoperazine has four binding sites on AGP, three of which had the same lower affinity (551). Three binding sites with the same affinity are found for ticlopidine and PCR 2362 (195). For dipyridamole at least two binding sites on AGP were found by El-Gamal et al. (157) using Scatchard plot analysis; according to these authors, the high affinity site is located in a hydrophobic part of the protein chain of AGP. For phenothiazine neuroleptics, there is one common binding site on AGP. For these compounds, structural parameters other than the lipophilicity determine the binding (158). Schley (460), studying the binding of phenothiazines, observed that the high affinity of these compounds was caused mainly by the hydrophobic phenothiazine structure itself. The amino acid sequence of the AGP molecule allows one to speculate that the

32

hydrophobic region of the phenothiazine derivative may interact with the region 21-31 of the amino acid sequence, and that the piperazine side chain may cause ionic interactions with glutamic acids 177 and 178 at the other end of the AGP molecule (461). However, the affinity of the phenothiazines was also influenced by other factors. These results support the conclusion that the interactions of phenothiazines with AGP are not exclusively hydrophobic. Mueller et al. (362-364) also found only one binding site on AGP for a series of psychotropic drugs. Recently Ferry et al. (160) studied the interaction between antidepressants and  $\alpha_1$ -adrenergic receptor antagonists as regards their binding to AGP. They found that the binding must be of a hydrophobic nature and that more than one binding site must be involved. Meuldermans et al. (348) reported on the hydrophobic nature of the binding of several analgesics (fentanyl, alfentanil, sulfentanil, and lafentanil) to AGP. Lemaire and Tillement (305) found a correlation between the partition coefficient and the drug binding of betablockers by performing drug displacement studies of pindolol. Their results indicate that the AGP binding with beta-blockers must be predominantly of a hydrophobic nature. From studies on displacement of lidocaine by basic drugs, Goolkasian et al. (197) concluded that on AGP there was only one nonspecific site.

Whereas the above-mentioned literature points to the hydrophobic nature of the binding of drugs to AGP, the decrease in the propranolol binding found after desialylation (582) points to an electrostatic interaction as well. Robert et al. (433) also observed a decrease in the nicergoline binding after desialylation, but only when followed by carboxymethylation of the AGP. Friedman et al. (177) reported recently that desialylation of AGP reduced the binding for chlorpromazine, but not for propranolol, progesterone, and dipyridamole. Drayer et al. (149) tried to describe a relation between the lipophilicity of the drugs and the binding to AGP for a series of antiarrhythmics and for a series of beta-blockers, respectively, but they concluded that factors other than lipophilicity are involved in the drug-protein interaction.

Schley and Mueller-Oerlinghausen (463) recently investigated the binding of various tricyclic neuroleptics and antidepressants to AGP. They could not find a correlation between the association constants of the investigated compounds and their antipsychotic potency or their ionization constant.

Busby and Ingham (97) recently suggested the use of fluorescent probes to study the interaction of AGP with other ligands.

Several studies (11, 28–30, 244, 540, 557) have reported that drug binding in plasma is stereoselective, especially in the case of basic drugs, due to stereoselective binding to AGP. This stereoselectivity of AGP is used nowadays in separation techniques for drugs which involve the use of a chiral AGP column (229–234, 402, 458, 459).

The role played by factors other than hydrophobic

forces may become evident when one studies the structure of the several basic compounds used in binding studies. Nearly all of them contain a structure consisting of a tertiary nitrogen, linked via a bridge to an aryl system. The bridge is 2 to 4 atoms large and is composed of carbon and nitrogen, mainly of the type C-C, C-C-C, C-C-C, C-C-N, and C-C-C-N. It is reasonable to assume that the aryl system contributes to hydrophobic forces, and that the positively charged tertiary nitrogen contributes to ionic forces.

## E. Factors Influencing the Characteristics of the Binding of Drugs to Alpha-1-acid Glycoprotein

In this section we shall discuss the influence of several factors, such as the experimental methods and conditions, the physical chemical properties of AGP, and the method used to interpret the drug binding to AGP. These factors are summarized in table 13. Some additional comments will be made.

Recently Yost and DeVane (592) observed a large diurnal variation in the AGP level in plasma of healthy volunteers, namely, a fluctuation of up to 49% within 24 h, which was sex related. This finding should be taken into account in the interpretation of binding data.

The use of Vacutainers for blood collection or the storage of blood samples in plastic containers can result in a decreased drug binding, due to the presence of plasticizers, such as tris(2-butoxyethyl)phosphate (table 13, no. I). These plasticizers were found to selectively displace basic drugs from their binding sites on AGP. This finding was supported by the fact that plasticizers had no effect on the drug binding in plasma deficient in AGP (412-415).

Experimental conditions such as pH, temperature, buffer composition, methods used for drug level determinations, and serum protein concentrations used need to be rigorously controlled because these factors influence the drug binding (table 13, nos. III-V).

The pH of the drug-containing plasma should be checked before and after dialysis (418). Drug binding in plasma increases with increasing pH (table 13, no. III), except in the case of the binding of alfentanil (348) which proved to be pH independent. It should be pointed out that, up until recently, no one has studied the possible effect of pH on the binding of drugs to isolated AGP. It should be noted that such a pH effect has been demonstrated clearly for HSA. Isolated HSA shows a pHdependent drug binding due to its conformation transition in the physiological pH range (150, 151, 189, 283, 386, 546, 574-576). The pH of serum can also increase during storage (357). Moreover, it is shown that, during storage of plasma, desialylation of AGP and loss of hexose from AGP can occur (522, 523).

The free fraction can be determined using several direct techniques such as equilibrium dialysis, ultrafiltration, high performance liquid chromatography, ultracentrifugation, and gel filtration (113, 130, 189, 210, 246,



PHARMACOLOGICAL REVIEWS

## DRUG BINDING TO HUMAN AGP IN HEALTH AND DISEASE

TABLE 13								
Factors influencing the characteristics of the binding of drugs to AGP								

No.	Factor	Effect	Drug	Ref.
I	Use of Vacutainers for	Drug displacement from	Quinidine	126, 143, 153,
	blood collection, or	AGP by plasticizers,	-	154, 170, 176,
	storage in plastic	resulting in increased		354, 387, 415,
	containers	free fraction of espe-		492
		cially basic drugs	Propranolol	<b>452</b> <b>44, 126, 415, 572</b>
		ciany basic urugs	-	
			Oxprenolol	44, 74
			Alprenolol	74, 407
			Pindolol	305
			Lidocaine	143, 197, 510
			Chlorpromazine	350
			Imipramine	75, 118, 255, 350
			Amitriptyline	118, 350, 415
			Nortriptyline	118
			Thioridazine	381
			Ketamine	135
			Diazepam	5
			-	
			Disopyramide	132
п	Presence of heparin	Heparin increases free	Lidocaine	209
		fraction of basic	Imipramine	289
		drugs; this effect can	Propranolol	449, 583, 584
		be suppressed by pro-		110,000,001
		tamine sulfate		
		camine surface		
ш	Experimental condi-	Increase of binding with	Neuroleptics	81a
	tions during free	increasing pH in	Beta-blockers	228
	fraction affect de-	plasma	Quinidine	
		plasma	•	87, 210
	termination in		Lidocaine	96, 189, 209, 343,
	plasma, such as pH,			418
	temperature, buffer,		Ketamine	135
	storage effect		Bupivacaine	140
			Erythromycin	142, 420
			Progesteron	182
			Etidocaine	357
			Verapamil	187, 220, 239
			Fentanyl	336, 348
			Sufentanil	348
			Lofentanil	348
			Imipramine	289
			Disopyramide	397, <b>495a</b>
			Thioridazine	381
			Phencyclidine	390
		Temperature effect can	Propranolol	398
		be different		
		Decrease of binding	Progesterone	181
		with increasing	Imipramine	289
		temperature	Propranolol	398
		Increase of binding	Erythromycin	420
		with increasing	Fentanyl	142, 239
		temperature		174, <i>4</i> 07
		No temperature ef-	Disopyramide	495a
		fect	2	1004
		Different buffer types	Quinidine	87, 207, 238
		and different ionic	Progesterone	182
		strength of the buf-	Lidocaine	209, 418
		fers can influence the	Fentanyl	239
		free drug level	Imipramine	239
		THE MINE IEVEL	-	
			Erythromycin	420
		Increased binding dur-	Etidocaine	357
		ing storage		
R7	Dama fastar !- A-	Values alife James J		40.05.00.140
IV	Some factors influ- ence when using	Volume shift, depend- ing on dialysis time,		40, 87, 93, 140, 189, 207, 227,

## TABLE 13—Continued

No.	Factor	Effect	Drug	Ref.
	dialysis time; the	tion of free fractions;		289, 292, 309,
	with drug spiked	dialysis time is		316, 322, 341,
	side; pH change	shorter when plasma		360, 379, 398,
	during equilibrium	or plasma protein		418, 460, 463,
	dialysis; adsorption	fraction side is spiked		495, 495a, 501,
	of drug; presence of	with drug; pH		539
	<b>•</b> ·•	change, adsorption of		
	organic solvents			
		drug, and the pres-		
		ence of organic sol-		
		vents during equilib-		
		rium dialysis can		
		change drug binding		
v	Method used for the	Different degrees of de-		12, 58, 489, 490
•	determination of	sialylation of AGP		,,,
	AGP concentration	result in a different		
	AGF concentration	AGP level		
		AGF level		
VI	Different origins of	Contamination of albu-	Disopyramide	223, 317
	plasma or AGP	min delipidation of	Progesterone	181, 565
	sample resulting in	AGP during isolation	Bupivacaine	128
	different degrees of	increases binding	Lidocaine	285
	desialylation of	Desialylation occurring	Liuvulliv	58, 73, <b>489</b>
	-	in several diseases		00, 10, 400
	AGP, different con-			
	tent of fatty acids,	affects binding dif-		
	different degree of	ferently	<b>.</b>	
	polymerization of	Desialylation of AGP	Propranolol	582
	AGP, occurrence of	decreases drug	Chlorpromazine	177
	abnormal AGP in	binding	Nicergoline	433
	plasma due to dis-	Desialylation of AGP	Propranolol, dipyridamole,	177
	ease states, or drug	has no effect on	progesterone	
	therapy	drug binding		
	••	Combination of carbox-	Nicergoline	433
		ymethylation and de-	-	
		sialvlation of AGP		
		decreases drug bind-		
		ing		
		Degree of polymeriza-	Progesterone	211, 212, 507, 508
		tion of AGP depends	Ū	
		on purification proce-		
		dure used and		
		changes the drug		
		binding	Deemethedinging	94
		Increased content of	Desmethylimipramine	84
		sialic acid due to		
		treatment with phe-		
		nobarbital increases		
		binding		
VII	Microheterogeneity	The different molecular	Not studied until now	15, 529; table 4,
	and variants of	weights used influ-	with exception of Tin-	section IID
	AGP	ence the magnitude	uely et al. (529)	
		of the binding param-		
		eters; the effect of		
		microheterogeneity		
		and the occurrence of		
		several AGP variants		
		· · · · ·		
		on drug binding		
	• • • • •	should be controlled		0 4 00 40 00
VIII	Interpretation of	Models with different	Drugs with their binding	2, 4, 22, 40, 93,
	binding data	classes of binding	parameters are already	166, 275, 276,
		sites and different	summarized in tables 7–	327, 337, 365,
		graphical methods	12	417, 422, 437,
		are used for the fit-		438, 457, 505,
		ting of the experi-		581
		mentally determined		
		data, resulting in dif-		
		ferent binding pa-		

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PHARMACOLOGICAL REVIEWS

PHARM REV

292, 394, 395, 398, 412, 414, 415, 463, 505, 517, 549). Parsons and Fan (394) recently reported the loss of propranolol during ultrafiltration by binding to the membrane and a larger continuous loss to the O-ring. Evaluation of the various methods by Kurz et al. (292) led these authors to the conclusion that the values achieved by equilibrium dialysis seem to come closest to the real extent of binding. This is the technique that is used most frequently in binding studies. However, several experimental conditions have to be considered (table 13, no. IV). During equilibrium dialysis a dilution of plasma can occur, due to osmotic equilibration; this will result in a relative overestimation of the free fraction. Levy et al. (309) discussed the dilution effect of plasma in cases where the drug binding was also dependent on the drug concentration. The volume shift that occurs increases with dialysis time (207, 243, 289, 539). Lohman et al. (322) reported recently that the relative overestimation of the free fraction, particularly with highly bound drugs, can be up to 60%, and they concluded that many published data may be incorrect. To attenuate this volume shift, several authors have proposed the addition of dextran to the buffer (316) and the use of competitive equilibrium dialysis (379). Dialysis time can be shortened when the drug is added to the plasma-containing side (341, 384). In order to shorten the dialysis time, Hwang and Bayne (247) proposed the use of a dynamic method for the estimation of the extent of plasma protein binding. Tozer et al. (539) proposed an equation which can be used to correct for the volume shift and for the change in the total drug concentration occurring during dialysis experiments in the plasma-containing compartment. When free fractions are calculated from equilibrium dialysis experiments, one should not use the initial total drug concentration as the total drug concentration, but instead one should use the drug concentration in the plasma compartment at equilibrium (40, 501, 539). One should check for adsorption to the dialysis chambers and the dialysis membranes (93, 360) and the effect on the drug binding of the presence of low concentrations of organic solvents (93, 460). Schley (460) reported that 1 to 5% ethanol decreased the perazine binding by about 12%, whereas adsorption was larger in the buffer-containing dialysis chamber.

Several methods can be used to determine the AGP level in plasma (table 13, no V). These include the radial immunodiffusion procedure of Manzini (12, 328, 329), the electroimmunodiffusion method of Laurell (12, 299), laser nephelometry (57, 416), solid-phase enzyme-linked immunosorbent assay (116, 148, 558), radioimmunoassay (RIA) (185), concanavalin A crossed immunoaffinoelectrophoresis (65–69, 373, 374, 563), fast protein liquid chromatography (536), immunoturbidimetric assay (138, 331), and a fluorimetric method using auramine O (516, 517). In most studies only one of the above-mentioned methods is used, and the various methods are hardly ever compared. Sugiyama et al. (516) found the same AGP concentration using auramine O or RIA, as did Haram et al. (218) using laser nephelometry or radial immunodiffusion. Bordas et al. (73, 74) reported, however, that the amounts of AGP estimated by electroimmunodiffusion in plasma were much lower than those actually present as assayed by radial immunodiffusion, the differences being due to different degrees of desialylation of AGP. The difference in the amounts of AGP revealed by the two immunological methods can be used as a basis for estimating the degree of sialylation of AGP; this can vary during several disease states, such as in liver disease and during inflammation (58, 73, 74, 489, 490). Too high or too low estimates of AGP values influence the binding parameters that are calculated.

Although the physical-chemical properties of AGP preparations can vary considerably, as already discussed in section II, the influence of these physical chemical properties on the drug binding is rarely determined in binding studies (table 13, nos. VI and VII). The binding of progesterone to AGP, being the first drug for which the affinity to AGP was described, proved to be dependent on the isolation method used for the AGP (table 11, no. XXXVIII). AGP isolated with an ethanolic precipitation method had a higher affinity for drugs than AGP isolated following a chromatographical procedure. AGP isolated chromatographically and treated with a mixture of alcohol-acetone had the same affinity to progesterone as AGP isolated with an ethanolic precipitation method. From these results it follows that AGP isolated with different methods can have a different degree of delipidation, which results in altered drug binding. Defatting has also an effect on the stability of AGP (213). The stability, denaturation, and degree of polymerization of AGP, as already discussed in section II, are dependent on the degree to which AGP is defatted and desialylated (212). Polymerization results in two types of polymers (507), each of which has a different biological activity (35, 388, 507) and a different binding to progesterone (508). Desialylation occurring during several isolation procedures for AGP (section II A), especially at low pH. also has an effect on the drug binding (table 11) and can be different for each drug. Wong and Hsia (582) found, for example, that desialylation reduced the propranolol binding, whereas the progesterone binding did not change. Robert et al. (433) found that desialylation had no effect on the nicergoline binding, whereas a combination of desialylation and carboxymethylation decreased the nicergoline binding. Carboxymethylation alone did not influence the nicergoline binding.

### V. Concluding Remarks

The literature on AGP has been reviewed up to 1987. From the literature, especially from the studies after 1980, the following conclusions can be drawn:

physical-chemical properties of AGP, such as the structure and the degree of desialylation of the carbo-

hydrate moiety, the degree of polymerization, the molecular weight, the microheterogeneity, and the binding parameters, can change due to disease states and due to the procedures used for its isolation; more attention should therefore be paid to the methodology of AGP preparation (section II);

AGP is indeed an acute phase protein, the plasma level of which can be used as a diagnostic and prognostic aid during clinical therapy; this has implications for the monitoring of the free fractions of basic drugs during clinical therapy (section III);

the large variations observed in the binding ratios of basic drugs in plasma during several physiological and pathological states are correlated with the large variations in the plasma level of AGP; this leads one to the conclusion that AGP plays an important role in the plasma binding of basic drugs (sections III and IV); and
 the experimentally determined characteristics of the binding of basic drugs to AGP are strongly dependent on experimental factors and on the physical-chemical properties of the AGP sample itself; therefore, more attention should be paid to the standardization of the experimental conditions and to the characterization of AGP (sections II to IV).

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PHARM REV 37

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